

# STOICHIOMETRY WORKSHEET

1. Calculate the number of moles of NaOH that are needed to react with 500 g of H<sub>2</sub>SO<sub>4</sub> according to the following equation:



2. Calculate the mass of NH<sub>3</sub> that can be produced from the reaction of 125 g of NCl<sub>3</sub> according to the following equation:



3. Identify the limiting reactant and determine the mass of CO<sub>2</sub> that can be produced from the reaction of 25 g of C<sub>3</sub>H<sub>8</sub> with 75 g of O<sub>2</sub> according to the following equation:



4. How many grams of SO<sub>2</sub> are produced when 152 g of CS<sub>2</sub> reacts with 48.0 g of O<sub>2</sub> according to the following equation:



# Chemistry Semester 1 Unit 9 Stoichiometry Answers

**Stephen DeMeo**



## **Chemistry Semester 1 Unit 9 Stoichiometry Answers:**

**Instructor's Manual** Michell J. Sienko, Robert A. Plane, 1966      Stanford University Bulletin Stanford University, 1991

**Mapping College Chemistry** Stephen DeMeo, 2019-03-01 This text is a chemistry problem solving resource appropriate for teachers and their students who are enrolled in high school Advanced Placement Chemistry or in a first year college General Chemistry course The book incorporates a chemistry problem solving plan one that uses an innovative graphic organizer strategy The strategy successfully evaluated with students combines problem solving processes with chemical concepts that will allow students to solve the most common and difficult problems encountered in the first year of chemistry Topical problem solving will focus on limiting reactant stoichiometry identifying types of chemical reactions equilibrium acid base equilibria and electrochemistry Why would this resource be of interest to chemistry students To be successful to get into a well known college medical school physical therapy or graduate program often requires that students get an A in your pre requisite Introductory General Chemistry course To make matters worse many college professors feel that only a few students should get A grades and therefore they give difficult exams that many students fail this is the weeding out process that every pre health student is apprehensive about To succeed in this competitive environment entails not just studying harder or longer it means re organizing textbook content so that it is meaningful to the student This is the first text of its kind to employ a reliable research based strategy that incorporates a decision based visual tool to solve chemistry textbook problems ones that can make or break a career      The Software Encyclopedia 2001 ,2001      *The Science Teacher* ,2008

**Process Oriented Guided Inquiry Learning (POGIL)** Richard Samuel Moog, 2008 POGIL is a student centered group learning pedagogy based on current learning theory This volume describes POGIL s theoretical basis its implementations in diverse environments and evaluation of student outcomes      *General Catalog - Texas Tech University* Texas Tech University, 1944      *College Credit Recommendations* ,1996      New Technical Books New York Public Library, 1970

*Agents of Change* Clarence Walter Mangham, 1995 Volume one is the first of a series of historical documents designed to provide information on instructional improvement projects that have received grants and loans from the California Community College Fund for Instructional Improvement      **Chemical Engineering Education** ,2004      Choice ,1974

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christian dior automne hiver 1949 1950 traduction par julie rodhon article initialement publié sur

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web apr 7 2020 curated by vogue editors hamish bowles and chloe malle this updated and expanded edition of vogue and the metropolitan museum of art costume institute covers five years of the met costume institute s exhibitions and galas through the lens of the influential style magazine

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approximately 250 items drawn from the costume institute s permanent collection some very rarely seen in public before will be displayed in an entirely new way max hollein the met s marina kellen french director and ceo said

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