

# Periodic Trends Worksheet

Using the periodic table and your knowledge of periodic trends, answer the following questions:

1 Circle the best answer that has the:

- |   |                            |    |    |    |    |    |
|---|----------------------------|----|----|----|----|----|
| a | smallest electron affinity | N  | P  | As | Sb | Bi |
| b | largest electronegativity  | Ga | Ge | As | Sc |    |
| c | largest ionization energy  | O  | S  | Sc | Te | Po |
| d | lowest ionization energy   | O  | S  | Se | Te | Po |
| e | smallest atomic radius     | F  | Cl | Br | I  | At |
| f | largest atomic radius      | Li | Na | K  | Rb | Cs |

2 Low ionization energy is a characteristic of a \_\_\_\_\_

3 What does atomic radius measure? \_\_\_\_\_

4 Describe the change in the atomic size across a period and down a group.

5 Which element has the largest atomic radius? \_\_\_\_\_

6 Explain the main difference between metals and nonmetals.

7 Determine if the listed pairs of elements have similar or different properties.

- |   |                      |   |                    |
|---|----------------------|---|--------------------|
| a | Potassium & Rubidium | b | Calcium & Barium   |
| c | Sodium & Chlorine    | d | Helium & Krypton   |
| e | Phosphorus & Oxygen  | f | Lithium & Fluorine |

8 Which element has the largest electronegativity? \_\_\_\_\_

9 Which group (name) would have the lowest ionization energies?

\_\_\_\_\_

10 Which group (name) has the highest ionization energies?

\_\_\_\_\_

# Chemistry Practice Test Periodic Trends And Orbitals

**RJ Shavelson**



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**Chemistry: The Easy Way** Joseph A. Mascetta, Mark Kernion, 2019-08-06 A self teaching guide for students Chemistry The Easy Way provides easy to follow lessons with comprehensive review and practice This edition features a brand new design and new content structure with illustrations and practice questions An essential resource for High school and college courses Virtual learning Learning pods Homeschooling Chemistry The Easy Way covers Atomic Structure Chemical Formulas Electrochemistry The Basics of Organic Chemistry And more

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**Chemistry Class 11** Dr. S C Rastogi., Er. Meera Goyal, 2022-06-14 This Book has been written in according with the New Syllabus of Madhyamik Shiksha Mandal Madhya Pradesh Bhopal based on the curriculum of CBSE NCERT Including solved questions of NCERT book based on new examination pattern and mark distribution Highly Useful for NEET AIIMS IIT JEE APJ AKTU and Engineering Medical Examinations Syllabus Unit I Some

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*Proceedings of the Robert A. Welch Foundation Conferences on Chemical Research* ,1970      **Chemistry of Organic Sulfur Compounds** ,1966      **The Transuranium Elements--the Mendeleev Centennial** ,1970      **Foundations of College Chemistry** Daniel Barker Murphy,Viateur Rousseau,1969

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