

Chemistry Notes 14.3

Three Basic Assumptions of the kinetic theory:

- 1) Gas is composed of particles, usually molecules or atoms
- 2) Particles in a gas move rapidly in constant random motion (they move in straight line paths only colliding with one another or other objects)
- 3) Collisions are perfectly elastic: kinetic energy is transferred without loss from one particle to another

Equation:  $P \times V = n \times R \times T$

$R = 8.31 \text{ (L} \times \text{kPa) / (mol} \times \text{K)}$

$n = (P \times V) / (R \times T)$

Ideal Gas:

- 1) A gas is composed of particles with a mass that is low enough to be considered almost zero.
  - But a real gas has limitations to the pressure it can be compressed to before it condenses into a liquid

Real gases differ from ideal gases in the following ways:

- Real gases have small attractive and repulsive forces between particles, and ideal gases do not
- Real gas particles have a volume, and ideal gas particles do not
- Real gas particles collide inelastically and lose energy with collisions, and ideal gas particles collide elastically
- Real gases are not infinitely compressible

Ideal gases don't exist because they do take up space, and there are attractive forces

Diffusion: molecules moving from areas of high concentration to low concentration

- Perfume

Effusion: Gas escaping through tiny hole in container

- Both of these depend on the molar mass of the particle, which determines the speed
- The rate of effusion and diffusion is inversely proportional to the square root of the molar mass

To solve Graham's law problems, put whatever you're looking for in the numerator  
6777

# Chemistry Ch 14 Guided Reading

**Antony C. Wilbraham, Dennis D.  
Staley, Michael S. Matta, Edward L.  
Waterman, Prentice-Hall Staff**

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**Chemistry** James N. Spencer,George M. Bodner,Lyman H. Rickard,2002-02-28  
The second edition of Spencer s Chemistry Structure and Dynamics has been the most successful reform project published for the General Chemistry course The authors have revised the text by building on the recommendations of the ACS s Task Force on the General Chemistry Curriculum and suggestions from the adopters of the first edition This innovative text provides a fifteen chapter introduction to the fundamental concepts of Chemistry A collection of additional topics at the end of each chapter allow instructors to supplement and tailor their courses according to individual need Three major themes link the content of the book the process of science the relationship between molecular structure and physical chemical properties and the relationship between the microscopic and macroscopic levels

Essential Biochemistry Charlotte W. Pratt,Kathleen Cornely,2021-03-23 Essential Biochemistry 5th Edition is comprised of biology pre med and allied health topics and presents a broad but not overwhelming base of biochemical coverage that focuses on the chemistry behind the biology This revised edition relates the chemical concepts that scaffold the biology of biochemistry providing practical knowledge as well as many problem solving opportunities to hone skills Key Concepts and Concept Review features help students to identify and review important takeaways in each section

**On the Origin of life** Dr. I. Pearson,2025-09-15 This book proposes a bold new answer to one of science s oldest questions Instead of starting with random chemistry On the Origin of Life shows how living systems could have begun with deterministic field structures built from protons electrons and neutrinos through mechanisms described by Stein Theory These simple geometric rules of spin alignment and corridor formation created stable feedback loops memory and responsiveness long before DNA proteins or cells existed The book traces life s journey from invisible neutrino based dark life through rock bound Darwinian chemistry and droplet proto cells into the rise of membranes DNA and organisms It demonstrates how many of these early processes remain embedded in modern biology and how understanding them opens new possibilities in medicine synthetic biology and the search for life in the universe

**Design and Analysis in Quantitative Educational Research** Kamden K. Strunk,Mwarumba Mwavita,2024-08-22 Design and Analysis in Quantitative Educational Research Univariate Designs in SPSS presents an integrated approach to learning about research design alongside statistical analysis concepts The authors maintain a focus on applied educational research

throughout the text with practical tips and advice on how to do high quality quantitative research This book teaches research design including epistemology research ethics forming research questions quantitative design sampling methodologies and design assumptions and introductory statistical concepts including descriptive statistics probability theory and sampling distributions basic statistical tests like z and t ANOVA designs including more advanced designs like the factorial ANOVA and mixed ANOVA simple linear regression and multiple regression using SPSS for analysis Designed specifically for an introductory graduate course in research design and statistical analysis the book takes students through principles by presenting case studies describing the research design principles at play in each study and then asking students to walk through the process of analyzing data that reproduce the published results The new edition has been updated to include the latest contemporary research developments coverage of IBM SPSS Statistics version 29 and also features new case studies many focused on race and racism as well as other equity and justice work and additional content on covariance and correlation An online eResource is also available with data sets This textbook is tailor made for first level doctoral courses in research design and analysis and will also be of interest to graduate students in education and educational research

Books in Print Supplement ,2002

**Focus on Earth Science** ,2001

**The Viologens** Paul M. S. Monk,1998

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