

## CHEMISTRY MIDTERM REVIEW

### 1.) CONVERSIONS IN METRIC UNITS

How many kilograms are in 143 grams?

Using KHDBDCM

Starting with Grams (g) – base unit. Moving to kilograms (kg); move the decimal three spaces to the right

143 g – 0.143 g

Using conversion factors

KNOW that kilo = 1000 base

$143 \text{ g} \times (1 \text{ kg}/1000\text{g}) = 0.143 \text{ kg}$

### 2.) CONVERSIONS WITH ENGLISH UNITS

How many milliliters are in 13.5 oz?

**MUST USE CONVERSION FACTORS (oz, lb, feet, inches...)**

Always start with what you know... Figure out where to go

13.5 oz

? mL

13.5 oz	1 cup	1 pt	1 qt	1 L	1000 mL
	8 oz	2 cups	2 pt	1.06 qt	1 L

= 398 mL are in 13.5 oz

Question: is ATM (atmospheres) a standard number?

Answer; No, it's a unit. There's not a particular number there. The gas constant has a number and a unit because it is a constant; atmospheres will differ depending on the problem.

Stoichiometry: Practice questions.

22.0 g of propane will combine with what mass (g) of O<sub>2</sub>?



# Chemistry Midterm Review Answers

**D Kirk**



## **Chemistry Midterm Review Answers:**

## **Chemistry Midterm Review Answers** Book Review: Unveiling the Power of Words

In some sort of driven by information and connectivity, the energy of words has been evident than ever. They have the capacity to inspire, provoke, and ignite change. Such may be the essence of the book **Chemistry Midterm Review Answers**, a literary masterpiece that delves deep to the significance of words and their impact on our lives. Compiled by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we will explore the book's key themes, examine its writing style, and analyze its overall effect on readers.

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