

Chemistry of Life Exercise**Objectives:**

1. Recognize and state the major characteristics of carbohydrates, proteins and lipids and their building blocks the monosaccharides, amino acids, glycerol and fatty acids.
2. Identify the main elements that compose the different classes of organic molecules and list how many covalent bonds one atom of each of the elements can form.
3. If given the structural formula for an organic compound, be able to identify all functional groups in the molecule and the type of compound (alcohol, aldehyde, amino acid, ketone or an organic acid including fatty acids).
4. Use common lab tests to test for the presence of simple sugars (reducing sugars), starch, lipids and proteins.
5. Know what substances are used to test for simple sugars (reducing sugars), starch, lipids and proteins and be able to recognize the positive results for each test.

Introduction:

Life, as we know it on earth, depends to a great extent on the structure, function and chemical reactivity of a diverse group of compounds including carbohydrates, proteins, lipids and nucleic acids. These common organic substances are major components of cells and/or fluids and secretions of the body. This exercise is intended to acquaint you with the basic structure and characteristics of these vital substances except for nucleic acids. Common lab tests will be used to identify simple sugars (reducing sugars), starch, lipids and proteins.

Partial list of Functional Groups of Organic Compounds:

Name	Abbreviation	Structure	Properties
Hydroxyl group (Alcohols)	-OH	-O-H	Polar
Carboxyl group (Amino acids, Organic acids)	-COOH	$\begin{array}{c} \text{O} \\ \parallel \\ -\text{C}-\text{OH} \end{array}$	Acidic
Amino group (Amino acids, Amines)	-NH ₂	$\begin{array}{c} \text{H} \\ \\ -\text{N}-\text{H} \end{array}$	Basic
Carbonyl group (Aldehydes)	-CHO	$\begin{array}{c} \text{H} \\ \\ -\text{C}=\text{O} \end{array}$	Polar
Carbonyl group (Ketones)	-C-CO-C-	$\begin{array}{c} \text{O} \\ \parallel \\ -\text{C}-\text{C}-\text{C}- \\ \\ \text{O} \end{array}$	Polar
Phosphate group	-H ₂ PO ₄	$\begin{array}{c} \text{OH} \\ \\ -\text{O}-\text{P}-\text{OH} \\ \\ \text{O} \end{array}$	Acidic

Chemistry Lab Answers

Frederick Hutton Getman



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Practical Chemistry Labs Leonard Saland,1989 Features self contained step by step activities using common materials and covering topics from food chemistry to papermaking and electrochemistry Illustrates the connection between the real world and chemistry concepts such as solutions chemistry acids and bases and more Includes teacher notes quizzes and answers to help monitor student progress *Chemistry in the Laboratory* James M. Postma,Julian L. Robert,J. Leland Hollenberg,2004-03-12 This clearly written class tested manual has long given students hands on experience covering all the essential topics in general chemistry Stand alone experiments provide all the background introduction necessary to work with any general chemistry text This revised edition offers new experiments and expanded information on applications to real world situations [Chemistry Lab Manual Answer Key, 2nd Edition](#) Bright Thinker,2024-07 **Problems and Problem Solving in Chemistry Education** Georgios Tsaparis,2021-05-17 Problem solving is central to the teaching and learning of chemistry at secondary tertiary and post tertiary levels of education opening to students and professional chemists alike a whole new world for analysing data looking for patterns and making deductions As an important higher order thinking skill problem solving also constitutes a major research field in science education Relevant education research is an ongoing process with recent developments occurring not only in the area of quantitative computational problems but also in qualitative problem solving The following situations are considered some general others with a focus on specific areas of chemistry quantitative problems qualitative reasoning metacognition and resource activation deconstructing the problem solving process an overview of the working memory hypothesis reasoning with the electron pushing formalism scaffolding organic synthesis skills spectroscopy for structural characterization in organic chemistry enzyme kinetics problem solving in the academic chemistry laboratory chemistry problem solving in context team based active learning technology for molecular representations IR spectra simulation and computational quantum chemistry tools The book concludes with methodological and epistemological issues in problem solving research and other perspectives in problem solving in chemistry With a foreword by George Bodner *Techniques in Organic Chemistry* Jerry R. Mohrig,Christina Noring Hammond,Paul F. Schatz,2006 Is the most comprehensive and detailed presentation of lab techniques available for organic chemistry students and the least expensive It combines specific instructions for 3 different kinds kinds of laboratory glassware and offers extensive coverage of spectroscopic techniques and a strong emphasis on safety issues **Engineer Update** ,1991

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A Laboratory Manual of Organic Chemistry for Beginners Arnold Frederick Holleman,1913
EJEL Volume 10 Issue 2 , *The English Catalogue of Books* Sampson Low,1906 Volumes for 1898 1968 include a directory of publishers *The Saskatchewan Bulletin* ,1962 **The United States Catalog** Mary Burnham,Carol Hurd,1928 **The Annual American Catalog** ,1904 **Laboratory Experiments for Chemistry, the Central Science, 5th Ed** John Henry Nelson,1991 *Outlines of Theoretical Chemistry* Frederick Hutton Getman,1913 **ECMLG 2011 Proceedings of the 7th European Conference on Management Leadership and Governance** Charles Despres,2011-06-10 **American Druggist and Pharmaceutical Record** ,1898 **Industrial and Engineering Chemistry** ,1923 **The Annual American Catalogue Cumulated 1900-1902** ,1903 *The Merck Report* ,1902

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