

CHAPTER # 1

STOICHIOMETRY

Q1. Define stoichiometry.

Ans: Stoichiometry:

The study of relative amounts of substances involved in a chemical reaction is called Stoichiometry. Such phenomenon is studied through the knowledge of Stoichiometry (Greek word *Stolcheion* means element and *metry* means measurement).

Importance of Stoichiometry:

Stoichiometry is essential when quantitative information about a chemical reaction is required. Moreover it is important to predict yields of chemical products.

Q2. Explain the significance of stoichiometry with the help of example.

OR

How will you explain law of conservation of mass in the case of combustion of hydrogen fuel in rockets?

Ans: Combustion of hydrogen fuel in rockets:

1. Consider a rocket manufacturer uses liquid hydrogen as a fuel. He may have to determine how much fuel is necessary for a particular flight. Hydrogen burns in oxygen (of air) to produce water.



It states that

- (i) Two moles of Hydrogen react with one mole of oxygen to form two moles of steam.
 - (ii) Two molecules of Hydrogen react with one molecule of oxygen to produce two molecules of steam.
 - (iii) Four grams of hydrogen react with thirty-two grams of oxygen to produce thirty-six grams of water. Here the total mass of reactants is equal to the total mass of products. Thus it confirms the Law of conservation of mass.
2. Another example is the reaction taking place in a gas barbecue. This is the example of combustion to form carbon dioxide and water. The balanced chemical reaction is



Q3. What do you understand by the term Mole?

Ans: Mole:

The atomic mass, formula mass and molecular mass of a substance expressed in grams is called Mole.

$$\text{Number of moles} = \frac{\text{mass in grams}}{\text{molecular mass}}$$

Example: One mole of O = 16 g

Chemistry Chapter 9 Stoichiometry

S Ashworth



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Adopting the Track of Appearance: An Psychological Symphony within **Chemistry Chapter 9 Stoichiometry**

In a world eaten by displays and the ceaseless chatter of fast connection, the melodic beauty and psychological symphony created by the published word often fade in to the background, eclipsed by the relentless noise and disruptions that permeate our lives. Nevertheless, located within the pages of **Chemistry Chapter 9 Stoichiometry** a charming fictional prize filled with fresh thoughts, lies an immersive symphony waiting to be embraced. Constructed by a wonderful composer of language, this fascinating masterpiece conducts readers on an emotional journey, well unraveling the concealed songs and profound affect resonating within each carefully constructed phrase. Within the depths of this touching assessment, we shall discover the book is key harmonies, analyze its enthralling writing model, and submit ourselves to the profound resonance that echoes in the depths of readers souls.

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