

CHEMICAL REACTIONS

webquest

emergency sub plans!

CHEMICAL REACTIONS Name: _____

Learn about chemical reactions by reading the text below.

1. What happens to matter in a chemical reaction?

2. Write the equation for a chemical reaction? Explain.

3. What happens to the particles and products in a rxn?

4. Draw the diagram of the collision and showing if it will be successful in the future?

5. What do you notice about the number of hydrogen atoms on each side of the reactants?

6. Complete the chart below with the four signs of rxn.

TYPE OF EVIDENCE	REACTIONS

CHEMICAL REACTIONS Name: _____

Learn about chemical reactions by reading the text below.

1. What is interesting about the composition of water?

2. Write a model of the composition of water in the form of a chemical equation?

3. What happens to the gases that are left over?

4. How does water in the air get there? (Chemical reaction or physical change?)

5. What is the relationship between oxygen and water?

6. What causes the clouds and condensation to form in a rxn?

7. How does water in the air get there? (Chemical reaction or physical change?)

8. Why does there not a lot of ammonia around?

9. What are the signs of a reaction? (Chemical or physical change?)

10. How do you know if the rxn is a chemical reaction?

11. Write a chemical equation for the reaction.

12. Explain what happens in a chemical reaction.

CHEMICAL REACTIONS Name: _____

Learn about chemical reactions by reading the text below.

1. Write a chemical equation for the reaction of hydrogen and oxygen to form water.

2. Write a chemical equation for the reaction of hydrogen and oxygen to form hydrogen peroxide.

3. Write a chemical equation for the reaction of hydrogen and oxygen to form water and hydrogen peroxide.

4. Write a chemical equation for the reaction of hydrogen and oxygen to form water and hydrogen peroxide.

5. Write a chemical equation for the reaction of hydrogen and oxygen to form water and hydrogen peroxide.

6. Write a chemical equation for the reaction of hydrogen and oxygen to form water and hydrogen peroxide.

7. Write a chemical equation for the reaction of hydrogen and oxygen to form water and hydrogen peroxide.

8. Write a chemical equation for the reaction of hydrogen and oxygen to form water and hydrogen peroxide.

9. Write a chemical equation for the reaction of hydrogen and oxygen to form water and hydrogen peroxide.

10. Write a chemical equation for the reaction of hydrogen and oxygen to form water and hydrogen peroxide.

CHEMICAL REACTIONS Name: _____

Learn about the signs of chemical reactions by reading the text below.

1. What are the signs of a chemical reaction?

2. Chemical reactions are all _____.

3. In a chemical reaction, the total mass of the reactants is _____ the total mass of the products.

4. In a chemical reaction, the total number of atoms of each element is _____ the total number of atoms of each element in the products.

5. Complete the chart with the number of oxygen atoms in each chemical equation. Then the relationship in your own words. (Using the data to make your own words.)

Chemical Equation	Reactants	Products
$2H_2 + O_2 \rightarrow 2H_2O$		
$2H_2 + O_2 \rightarrow 2H_2O_2$		
$2H_2 + O_2 \rightarrow 2H_2O + O_2$		

6. Complete the chart below with the number of oxygen atoms in each chemical equation. Then the relationship in your own words. (Using the data to make your own words.)

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Anne O'Daly



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