

## Chapter Test A

### Chapter: Chemical Equations and Reactions

Do the work provided below the letter of the item or phrase that best completes each statement or best answers each question.

1. The mass number of an atom is the sum of the number of \_\_\_\_\_ and the number of \_\_\_\_\_ in the nucleus of the atom.
  - a. electrons; neutrons
  - b. protons; neutrons
  - c. protons; electrons
  - d. neutrons; electrons
2. In the reaction described by a chemical equation, \_\_\_\_\_ is \_\_\_\_\_ and \_\_\_\_\_ is \_\_\_\_\_.
  - a. oxidized; hydrogenation; a product
  - b. oxidized; iron; product
  - c. reduced; iron; reactant
  - d. reduced; iron; product
3. Which of the following relationships is always true for a chemical reaction?
  - a. The products of a reaction are more stable than the reactants.
  - b. All of the reactants are used.
  - c. The products of a reaction are more stable than the reactants.
  - d. The reactants of a reaction are more stable than the products.
4. There is a negative enthalpy change for a reaction if the products \_\_\_\_\_ the reactants.
  - a. have a lower enthalpy than
  - b. are less stable than
  - c. have a higher enthalpy than
  - d. are more stable than
5. In a reaction, a chemical equation \_\_\_\_\_ the \_\_\_\_\_ of the reaction.
  - a. shows the enthalpy
  - b. shows the number of products
  - c. shows the number of reactants
  - d. shows the coefficients

# Chapter 9 Modern Chemistry Test

**Lingsheng Yao**



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