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ether solvent and the brown cloudy mixture was filtered by gravity filtration with fluted fast flow filter paper in a powder funnel into a 250 ml round bottom flask rbf leaving light brown residue and beige filtrate diethyl ether was

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web trimyristin in the major triglyceride found in nutmeg representing 20 25 of the dried weight of ground nutmeg to extract the trimyristin we performed a solid liquid extraction using the organic solvent methylene chloride CH_2Cl_2

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web isolation of trimyristin from nutmeg purpose the purpose of this experiment is to isolate myristic acid from ground nutmeg seeds nutmeg seeds are extracted using CH_2Cl_2 to yield crude trimyristin then recrystallized to yield pure trimyristin which will be hydrolyzed and undergo esterification

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web read standard reflux on page 201 of the organic chem lab survival manual prelab question 1 trimyristin is a triglyceride what are triglycerides prelab question 2 how do microwaves heat microwave procedure add approximately 2.5 g of ground nutmeg record exact mass and 15 ml of diethyl ether to microwave vessel

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web in this week's lab experiment the purpose of this lab is to new techniques isolate the chemical trimyristin from the common spice nutmeg this lab requires five techniques simple distillation vacuum filtration melting point determination reflux and extraction

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web hatice kocataş bursa uludağ university faculty of arts and sciences department of chemistry 16059 bursa turkey abstract in the present work a surfactant assisted convenient extraction method was developed for the isolation of trimyristin from nutmeg

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web trimyristin is found in the fixed oil of nutmeg the fixed oil comprises approximately 24 40 of the nutmeg seed trimyristin comprises 73 of the fixed oil overall trimyristin should have percent recovery of 18 29 1 figure 1 shows how trimyristin is triester formed from the dehydration reaction between glycerol and myristic acid oh

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web purpose the purpose of this lab is to extract trimyristin from nutmeg as well as synthesizing myristic acid from the extracted trimyristin using base hydrolysis and naoh acidification using hcl is necessary to characterize the product by finding the melting point

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web in this lab trimyristin is extracted from ground nutmeg occurring in many vegetable oils as well as myristic acid trimyristin consists of long saturated hydrocarbon chains and it is relatively nonpolar thus using a relatively nonpolar solvent trimyristin can be easily extracted by the concept of like dissolves like

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