

## Physical Science Chapter 6 Study Guide

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- **Atomic Theory of Matter**
  - Every element consists of tiny particles called atoms
  - All atoms of a particular element have the same properties
  - Different elements have different properties because their atoms are different
  - Atoms of different elements can combine in specific ways to form compounds
  - Chemical processes are the result of the rearrangement, combination, or separation of atoms
- **Abbreviation given to each element –chemical symbol—Know the bold print chemical symbols**
- **Parts of an atom**
  - **Nucleus—dense central core composed of protons and neutrons**
  - **Electrons surround the nucleus**
  - **Protons and neutrons are made up of small particles called quarks**
  - **Protons have positive charge and electrons have a negative charge**
- **Isotopes—atoms that are of the same element but have different number of neutrons**
- **Electron shells—regions around the nucleus that represent the energy level of electrons**
- **Ions—atoms with a charge**
  - **Cation—atom with positive charge**
  - **Anion—atom with negative charge**

# Chapter 6 Study Guide Physics

**Raymond A. Serway**



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Physics, Study Guide Sharon Ferrett, 1994-10      Study Guide to Accompany University Physics Harris Benson, 1991

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Stanley A. Williams, 1978      Physics, , Student Study Guide John D. Cutnell, Kenneth W. Johnson, 1997-09-04 Describes applications in medicine automobile features transportation home entertainment athletics household applications information processing detection devices camera technology and many more Contains numerous discussions and examples that focus on human physiology including muscle forces blood pressure the refraction of light by the eye and many others      **Study**

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