

UNIT 10: PERIODIC TABLE **PERIODIC TABLE** **PERIODIC TABLE** **PERIODIC TABLE** **PERIODIC TABLE**

Section 10.1 *(continued)*

In your notebook, read aloud the modern periodic table.

Copy the letterheads in the boxes on the left below from the periodic table to complete the table on the right.

| |
|-------|
| _____ |
| _____ |
| _____ |
| _____ |
| _____ |

| | |
|------------------------|-------|
| atomic mass | _____ |
| atomic number | _____ |
| electron configuration | _____ |
| chemical name | _____ |
| chemical symbol | _____ |

Put each item in Column A, under the letter of the matching item in Column B.

Column A

Column B

- | | |
|---|----------------------------------|
| _____ 18. A column in the periodic table | _____ a. metals |
| _____ 19. A row in the periodic table | _____ b. group |
| _____ 20. Group 1 elements | _____ c. period |
| _____ 21. Elements that are shiny and conduct electricity | _____ d. representative elements |
| _____ 22. Group 8 elements | _____ e. transition elements |

In the space on the left, write each of the statements in terms of the statement in bold. Change the underlined word or phrase to make it true.

- _____ 23. There are two representative elements of platinum.
- _____ 24. Most non-metal elements of the elements in the periodic table are metalloids.
- _____ 25. Group 1A elements except for hydrogen are located on the left side.
- _____ 26. Group 2A elements are the alkaline earth metals.
- _____ 27. Group 7A elements are highly reactive nonmetalic elements in halogens.
- _____ 28. Group 8A elements are inert nonmetallic elements known as noble gases.
- _____ 29. Metalloids have properties of both metals and non-metallic elements.

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M Tight



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