

# CODING MANUAL



# Coding Manual Manual

**R Barnett**



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web together the number of protons and the number of neutrons determine an element s mass number mass number protons neutrons if you want to calculate how many neutrons an atom has you can simply subtract the number of protons or atomic number from the mass number a property closely related to an atom s mass number is its atomic

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web the easiest way to find the atomic number is to look on a periodic table the atomic number is in the upper left corner or is the largest number on the square finding the number of protons the number of protons in an atom is equal to the atomic number of the element for example let s use oxygen

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