

## Periodic Table Trend Activities:



**Purpose:** The Periodic Table is a useful tool that helps us predict properties of various elements. In this activity, we will look at 3 properties:

1. **Atomic radius:** The distance from the atomic nucleus to the outmost occupied orbital in a atom
2. **Electronegativity:** The ability of an atom to attract a bonded pair of electrons from another atom
3. **Ionization Energy:** The energy needed to remove an electron from a neutral gaseous atom
4. **Electron Affinity:** The energy released when an electron is added to a neutral gaseous atom

**Procedure:** Click on the [Cool Periodic Table link](#) (click on the element symbol). Record the atomic radius, electronegativity, and first ionization energy for each element in the table below:

<http://www.lyncsburg.net/hhs/chemistry/trends/>

**Data:**

Element	Atomic Radius	Electronegativity	1 <sup>st</sup> Ionization Energy
Li			
Be			
B			
Na			
Mg			
Al			
C			
Si			
N			
P			

**Discussion:**

1. What appears to be the trend in atomic radius as you move from left to right in a row?
2. What appears to be the trend in atomic radius as you move down a column?

# Chemistry Lab Periodic Trends Computer Activity Answers

**Daniel F McAuley**



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## **Table of Contents Chemistry Lab Periodic Trends Computer Activity Answers**

1. Understanding the eBook Chemistry Lab Periodic Trends Computer Activity Answers
  - The Rise of Digital Reading Chemistry Lab Periodic Trends Computer Activity Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Lab Periodic Trends Computer Activity Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chemistry Lab Periodic Trends Computer Activity Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Lab Periodic Trends Computer Activity Answers
  - Personalized Recommendations
  - Chemistry Lab Periodic Trends Computer Activity Answers User Reviews and Ratings
  - Chemistry Lab Periodic Trends Computer Activity Answers and Bestseller Lists
5. Accessing Chemistry Lab Periodic Trends Computer Activity Answers Free and Paid eBooks
  - Chemistry Lab Periodic Trends Computer Activity Answers Public Domain eBooks
  - Chemistry Lab Periodic Trends Computer Activity Answers eBook Subscription Services
  - Chemistry Lab Periodic Trends Computer Activity Answers Budget-Friendly Options
6. Navigating Chemistry Lab Periodic Trends Computer Activity Answers eBook Formats

- ePub, PDF, MOBI, and More
  - Chemistry Lab Periodic Trends Computer Activity Answers Compatibility with Devices
  - Chemistry Lab Periodic Trends Computer Activity Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
    - Adjustable Fonts and Text Sizes of Chemistry Lab Periodic Trends Computer Activity Answers
    - Highlighting and Note-Taking Chemistry Lab Periodic Trends Computer Activity Answers
    - Interactive Elements Chemistry Lab Periodic Trends Computer Activity Answers
  8. Staying Engaged with Chemistry Lab Periodic Trends Computer Activity Answers
    - Joining Online Reading Communities
    - Participating in Virtual Book Clubs
    - Following Authors and Publishers Chemistry Lab Periodic Trends Computer Activity Answers
  9. Balancing eBooks and Physical Books Chemistry Lab Periodic Trends Computer Activity Answers
    - Benefits of a Digital Library
    - Creating a Diverse Reading Collection Chemistry Lab Periodic Trends Computer Activity Answers
  10. Overcoming Reading Challenges
    - Dealing with Digital Eye Strain
    - Minimizing Distractions
    - Managing Screen Time
  11. Cultivating a Reading Routine Chemistry Lab Periodic Trends Computer Activity Answers
    - Setting Reading Goals Chemistry Lab Periodic Trends Computer Activity Answers
    - Carving Out Dedicated Reading Time
  12. Sourcing Reliable Information of Chemistry Lab Periodic Trends Computer Activity Answers
    - Fact-Checking eBook Content of Chemistry Lab Periodic Trends Computer Activity Answers
    - Distinguishing Credible Sources
  13. Promoting Lifelong Learning
    - Utilizing eBooks for Skill Development
    - Exploring Educational eBooks
  14. Embracing eBook Trends
    - Integration of Multimedia Elements
    - Interactive and Gamified eBooks

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