



THOMAS'

# CALCULUS

*Thirteenth Edition*

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# Thomas Calculus Multivariable

**KJ Lindholm-Leary**



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web solution  $v_1 = 250$  l  $v_2 = 303$  k  $t_1 = 273$  k  $t_2 = 343$  k we will be applying charles law as pressure is constant here  $v_1 t_1 = v_2 t_2$  or  $v_2 = v_1 t_1 / t_2 = 250 \times 273 / 343 = 198.5$  l at  $90^\circ\text{C}$  a helium sample has a volume of 500 ml determine the temperature at which the volume of the liquid will become 240 ml

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