

ACIDS vs. BASES

On a **scale from 0 to 14**, lower pH solutions are **strong acids**; higher pH solutions are **bases**. Solutions measuring pH 7 are **neutral**.



Acid is a chemical substance that has a pH lower than seven.

An **acid** is any hydrogen-containing substance that is capable of donating a proton to another substance.

An **example of acid** substance can be found in the lemon juice which has a pH 2 level.



A **base** is a chemical substance that has a pH higher than seven.

A **base** is a molecule or ion able to accept a hydrogen ion from an acid.

An **example of base** substance can be found in the bleach which has a pH 12.6 level.

Acids Bases And Ph

**Great Britain. Foot-and-Mouth Disease
Research Committee**

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Kolthoff,1966 Technical Bulletin ,1917 *Chemistry and Our Universe* , Now turn to acids and bases Review the search for the defining qualities of these ubiquitous substances a quest that eluded scientists until independent discoveries made by J N Bronsted and T M Lowry in the 1920s Then hear how chemist Soren Sorensen devised the pH scale for measuring acidity and basicity *Acid-base Balance; Chemistry, Physiology, Pathophysiology* Arthur Gorman Hills,1973 Special Bulletin ... Michigan State University. Agricultural Experiment Station,1922 *Fluids, Electrolytes, and Acid Bases* George Tom Shires,1988 Designed for the use of critical care specialists anaesthetists and surgeons this text describes the general principles of fluid electrolyte and acid base balance followed by a discussion of the disturbance of balance in the critical care patient and in other specific abnormalities *Convention* National Electric Light Association. Convention,1929 **The Journal of the American Leather Chemists Association** American Leather Chemists Association,1922 **Fluid, Electrolyte, and Acid-base Disorders** Allen Ives Arieff,Ralph A. DeFronzo,1985 Progress Report of the Foot-and-Mouth Disease Research Committee Great Britain. Foot-and-Mouth Disease Research Committee,1925 Pocket Guide to Fluid, Electrolyte, and Acid-base Balance Mima M. Horne,1993

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