

# CHEMISTRY



**Atomic Structure**

Atoms are the smallest particles that cannot be created or destroyed in a chemical reaction. They are made up of three sub-particles: electrons, protons, and neutrons.

**Atomic Number**

The atomic number of an element is the number of protons present in the nucleus of an atom of that element. It is denoted by the symbol 'Z'.

**Atomic Mass**

The atomic mass of an element is the sum of the number of protons and neutrons present in the nucleus of an atom of that element. It is denoted by the symbol 'A'.

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Element	Atomic Number (Z)	Atomic Mass (A)
Hydrogen	1	1
Helium	2	4
Lithium	3	7
Boron	5	11
Carbon	6	12
Nitrogen	7	14
Oxygen	8	16
Fluorine	9	19
Neon	10	20

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