

Name Mel

Locker # \_\_\_\_\_

**Bonding & Molecular Structure****LEWIS STRUCTURES**

Indicate the total # of VALENCE electrons for each molecule (VE's). Calculate the number of bonded electrons from the skeleton (BE's). Calculate the Remaining Electrons (RE's = VE's - BE's) and distribute the remaining electrons onto the skeletal structure. Finally, satisfy the octets by creating double and triple bonds. Your final structures should be given below and your scratch paper should clearly show your work.

$\text{F}_2$ VE's = <u>14</u> BE's = <u>2</u> RE's = <u>12</u> $\text{:}\ddot{\text{F}}-\ddot{\text{F}}\text{:}$	$\text{H}_2$ VE's = <u>2</u> BE's = <u>2</u> RE's = <u>0</u> $\text{H}-\text{H}$	$\text{HF}$ VE's = <u>8</u> BE's = <u>2</u> RE's = <u>6</u> $\text{H}-\ddot{\text{F}}\text{:}$	$\text{NH}_3$ VE's = <u>8</u> BE's = <u>6</u> RE's = <u>2</u> $\begin{array}{c} \text{H} \\   \\ \text{H}-\text{N}-\text{H} \\   \\ \text{H} \end{array}$
Shape: <u>linear</u>	Shape: <u>linear</u>	Shape: <u>linear</u>	Shape: <u>trigonal pyramidal</u>
$\text{CH}_4$ VE's = <u>8</u> BE's = <u>8</u> RE's = <u>0</u> $\begin{array}{c} \text{H} \\   \\ \text{H}-\text{C}-\text{H} \\   \\ \text{H} \end{array}$	$\text{C}_2\text{H}_6$ VE's = <u>14</u> BE's = <u>14</u> RE's = <u>0</u> $\begin{array}{c} \text{H} \quad \text{H} \\   \quad   \\ \text{H}-\text{C}-\text{C}-\text{H} \\   \quad   \\ \text{H} \quad \text{H} \end{array}$	$\text{AlH}_3$ VE's = <u>6</u> BE's = <u>6</u> RE's = <u>0</u> $\begin{array}{c} \text{H}-\text{Al}-\text{H} \\   \\ \text{H} \end{array}$	$\text{BH}_3$ VE's = <u>6</u> BE's = <u>6</u> RE's = <u>0</u> $\begin{array}{c} \text{H} \\   \\ \text{H}-\text{B}-\text{H} \end{array}$
Shape: <u>tetrahedral</u>	Shape: <u>tetrahedral (both)</u>	Shape: <u>trigonal planar</u>	Shape: <u>trigonal planar</u>
$\text{C}_2\text{H}_4$ VE's = <u>12</u> BE's = <u>12</u> RE's = <u>0</u> $\begin{array}{c} \text{H} \quad \text{H} \\ \diagdown \quad / \\ \text{C}=\text{C} \\ / \quad \diagdown \\ \text{H} \quad \text{H} \end{array}$	$\text{C}_2\text{F}_4$ VE's = <u>36</u> BE's = <u>12</u> RE's = <u>24</u> $\begin{array}{c} \text{:}\ddot{\text{F}}\text{:} \quad \text{:}\ddot{\text{F}}\text{:} \\ \diagdown \quad / \\ \text{C}=\text{C} \\ / \quad \diagdown \\ \text{:}\ddot{\text{F}}\text{:} \quad \text{:}\ddot{\text{F}}\text{:} \end{array}$	$\text{CO}$ VE's = <u>10</u> BE's = <u>6</u> RE's = <u>4</u> $\text{:}\ddot{\text{C}}\equiv\ddot{\text{O}}\text{:}$	$\text{O}_2$ VE's = <u>12</u> BE's = <u>4</u> RE's = <u>8</u> $\text{:}\ddot{\text{O}}=\ddot{\text{O}}\text{:}$
Shape around one C: <u>trigonal planar</u>	Shape around one C: <u>trigonal planar</u>	Shape: <u>linear</u>	Shape: <u>linear</u>

# Lewis Dot Structure Work Sheet Answer Key

**RJ Shavelson**



**Lewis Dot Structure Work Sheet Answer Key:**

## Reviewing **Lewis Dot Structure Work Sheet Answer Key**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Lewis Dot Structure Work Sheet Answer Key**," an enthralling opus penned by a very acclaimed wordsmith, readers set about an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve in to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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