

Oxidation-Reduction Titrations I: Determination of Oxalate

B. Standardization of KMnO_4

	Trial 1	Trial 2	Trial 3
Mass of $\text{Na}_2\text{C}_2\text{O}_4$			
Weighing bottle initial mass	<u>2.4331</u>	<u>2.4331</u>	<u>2.4331</u>
Weighing bottle final mass	<u>2.6334</u>	<u>2.6358</u>	<u>2.6340</u>
Mass of $\text{Na}_2\text{C}_2\text{O}_4$	<u>0.2003</u>	<u>0.2027</u>	<u>0.2019</u>
Titration volume of KMnO_4			
Final reading	<u>26.4 mL</u>	<u>25.8 mL</u>	<u>16.5 mL</u>
Initial reading	<u>0.00 mL</u>	<u>0.00 mL</u>	<u>0.00 mL</u>
Volume of KMnO_4	<u>26.4 mL</u>	<u>25.8 mL</u>	<u>16.5 mL</u>

Calculations

mMoles of $\text{Na}_2\text{C}_2\text{O}_4$	_____	_____	_____
Volume of KMnO_4	_____	_____	_____

mMoles of KMnO_4	_____	_____	_____
Molarity of KMnO_4	_____	_____	_____

Average molarity _____

Standard deviation (show calculations) _____

Chemfax Lab Answers Oxidation Reduction Titrations

Patrick Vollmar



Chemfax Lab Answers Oxidation Reduction Titrations:

Oxidation-reduction Titrations in Acetic Acid as a Solvent Jacob David Neuss,1931 Titration Methods I.M. Kolthoff,R. Belcher,1957 Principles of Oxidation and Reduction A. G. Sharpe,1973 **Titration Methods:**
Oxidation-reduction Reactions ,1947 *The Application of Conductometric Methods to Oxidation-reduction Titrations*
Paul H. L. Walter,1956 **Titration Methods** I.M. Kolthoff,R. Belcher,1957 **Volumetric Analysis** Izaak Maurits
Kolthoff,Ronald Belcher,1957 **Volumetric Analysis** ,1942 **Oxidation Reduction Reactions in Aqueous Solutions**
Stanley Duckworth,1962

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