

## Chapter 9: Chemical Reactions

- \_\_\_\_\_ 1. Which of the following could be an indication of a chemical reaction of a compound or element?
- a. It is a solid at room temperature.
  - b. It melts at 631<sup>o</sup>C.
  - c. It expands upon freezing.
  - d. It burns in an atmosphere of chlorine.
- \_\_\_\_\_ 2. In this unbalanced chemical reaction, which components are the reactants?
- $$\text{C}_7\text{H}_{16} + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$$
- a.  $\text{C}_7\text{H}_{16}$  and  $\text{O}_2$
  - b.  $\text{C}_7\text{H}_{16}$  and  $\text{CO}_2$
  - c.  $\text{C}_7\text{H}_{16}$  and  $\text{H}_2\text{O}$
  - d.  $\text{CO}_2$  and  $\text{H}_2\text{O}$
- \_\_\_\_\_ 3. When  $\text{Ca}(\text{OH})_2$  reacts with  $\text{H}_2\text{SO}_4$ , what are the products.
- a.  $\text{Ca}(\text{OH})_2$  and  $\text{H}_2\text{SO}_4$
  - b.  $\text{CaH}_2$  and  $\text{H}_2\text{SO}_4$
  - c.  $\text{CaSO}_4$  and  $\text{H}_2\text{O}$
  - d.  $\text{CaO}$  and  $\text{H}_2\text{SO}_4$
- Hint: This is a double-replacement reaction, positive ions switch places.
- \_\_\_\_\_ 4. What are the products of the reaction between barium hydroxide and hydrochloric acid?
- a. Barium chloride and water
  - b. Barium hydroxide and water
  - c. Barium chloride and barium hydride
  - d. Chloric acid and barium hydroxide
- Hint: Write out the skeleton equation. This is a double-replacement reaction.
- \_\_\_\_\_ 5. This symbol separates the reactants from products and means "to yield".
- a. +
  - b. (s)
  - c.  $\rightarrow$
  - d. (g)
- \_\_\_\_\_ 6. In a chemical equation, the number written in front of the chemical formula is the
- a. Reactant
  - b. Subscript
  - c. Equation
  - d. Coefficient
- \_\_\_\_\_ 7. When there are exactly the same number and type of atoms on both sides of a chemical equation, the equation is \_\_\_\_\_.
- a. Balanced
  - b. Ionic
  - c. Synthetic
  - d. Decomposed
- \_\_\_\_\_ 8. The chemical equation  $\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$  is classified as a
- a. Synthesis reaction
  - b. Decomposition reaction
  - c. Single-replacement reaction
  - d. Double-replacement reaction
- \_\_\_\_\_ 9. In the combustion reaction of propane ( $\text{C}_3\text{H}_8$ ) with oxygen gas, what are the products?
- a.  $\text{C}_3\text{H}_8$  and  $\text{O}_2$
  - b.  $\text{C}_3\text{H}_8$  and  $\text{CO}_2$
  - c.  $\text{O}_2$  and  $\text{H}_2\text{O}$
  - d.  $\text{H}_2\text{O}$  and  $\text{CO}_2$
- \_\_\_\_\_ 10. The major physical indicator(s) of a combustion reaction is

# Chapter 9 Chemical Reactions Test

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