

Science

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(Chapter - 1) (Chemical Reactions and Equations) (Exercises) (Class 10)

Question 14:

In the refining of silver, the recovery of silver from silver nitrate solution involved displacement by copper metal. Write down the reaction involved.

Answer 14:



Question 15:

What do you mean by a precipitation reaction? Explain by giving examples.

Answer 15:

A reaction in which an insoluble solid (called precipitate) is formed is called a precipitation reaction.



In this reaction, barium sulphate is obtained as a precipitate. Hence, it is a precipitation reaction.

Question 16:

Explain the following in terms of gain or loss of oxygen with two examples each.

(a) Oxidation

(b) Reduction

Answer 16:

(a) Oxidation: Oxidation is the gain of oxygen.



Here, carbon (C) is oxidised to carbon dioxide (CO₂) and copper (Cu) is oxidised to copper oxide (CuO).

(b) Reduction: Reduction is the loss of oxygen.



Here, carbon dioxide (CO₂) is reduced to carbon monoxide (CO) and copper oxide (CuO) is reduced to copper (Cu).

Question 17:

A shiny brown coloured element 'X' on heating in air becomes black in colour. Name the element 'X' and the black coloured compound formed.

Answer 17:

The element 'X' is copper (Cu) which is converted into black coloured copper oxide (CuO).



Question 18:

Why do we apply paint on iron articles?

Answer 18:

Iron articles are painted because applying paints prevents them from rusting. When painted, the contact of iron articles from moisture and air is cut off. Hence, rusting is prevented their presence is essential for rusting to take place.

Chapter 9 Chemical Reactions Answers

Kathleen Armour



Chapter 9 Chemical Reactions Answers:

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Chemical Reactions and Chemical Reactors George W. Roberts, 2008-03-14 Focused on the undergraduate audience Chemical Reaction Engineering provides students with complete coverage of the fundamentals including in depth coverage of chemical kinetics By introducing heterogeneous catalysis early in the book the text gives students the knowledge they need to solve real chemistry and industrial problems An emphasis on problem solving and numerical techniques ensures students learn and practice the skills they will need later on whether for industry or graduate work

Study Guide with Answers to Selected Problems H. Stephen Stoker, Danny V. White, 2003-04 This supplement includes for each chapter a brief overview activities and practice problems to reinforce skills and a practice test The answers section includes answers for all odd numbered end of chapter exercises

Computer Modeling of Chemical Reactions in Enzymes and Solutions Arieh Warshel, 1997-03-28 This practical reference explores computer modeling of enzyme reactions techniques that help chemists biochemists and pharmaceutical researchers understand drug and enzyme action

Stochastic and Dynamic Views of Chemical Reaction Kinetics in Solutions A. M. Kuznetsov (dokt. fiz.-mat. nauk.), Alexander M. Kuznetsov, 1999-01-01

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approaching new situations methodically based on a repertoire of skills These skills are vital for successful problem solving in organic chemistry

Solutions Manual to Accompany General Chemistry Raymond Chang,1986

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Monograph Series ,1928

Soluble Silicates in Industry James Garrett Vail,1928

McGraw-Hill's SAT Subject Test: Chemistry, 2ed Thomas Evangelist,2009-02-01 We want to help you score high on the SAT Chemistry test We ve put all of our proven expertise into McGraw Hill s SAT Subject Test Chemistry to make sure you re fully prepared for this difficult exam With this book you ll get essential skill building techniques and strategies created by a leading high school chemistry teacher You ll also get 5 full length practice tests hundreds of sample questions and all the facts about the current exam With McGraw Hill s SAT Subject Test Chemistry we ll guide you step by step through your preparation program and give you the tools you need to succeed 5 full length sample tests with complete explanations for every question 40 top test items to remember on exam day A step by step review of all topics covered on the exam Teacher recommended tips and strategies to help you raise your score

Student Solutions Manual, Chemistry, Principles & Practice, Second Edition John M. DeKorte,1997

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The Lixiviation of Silver-ores with Hyposulphite Solutions Carl August Stetefeldt,1888

Chemistry Arnold B. Loebel,1978

Student Solutions Manual to Accompany Chemistry William Joseph Pietro,1990

General, Organic, and Biochemistry William Henry Brown,Elizabeth P. Rogers,1980

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