

Chapter 4 Chemical Bonds

Word Wise

Use the terms from the following list to fit each of the clues given below.

| | | |
|-----------|----------|------------|
| molecular | covalent | polyatomic |
| crystals | anion | polar |
| ionic | alloy | metallic |
| formula | bond | cation |

Clues

Vocabulary Terms

- A type of bond that holds cations and anions together _____
- A type of bond in which two atoms share a pair of valence electrons _____
- A neutral group of atoms that are joined together by one or more covalent bonds _____
- A term describing a covalent bond in which electrons are not shared equally _____
- An ion that contains a covalently bonded group of atoms _____
- An ion with a negative charge _____
- An ion with a positive charge _____
- A notation that shows what elements a compound contains and the ratio of the atoms or ions of these elements in the compound _____
- Solids whose particles are arranged in a lattice structure _____
- A mixture of two or more elements, at least one of which is a metal _____
- A type of bond that exists between a metal cation and the shared electrons that surround it _____
- The force that holds atoms or ions together _____

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Christian Drosten



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