

The Periodic Law

Section Review

Section 10.1: Answer the following questions in the space provided.

- Classify the central atoms in the following molecules as Lewis acids or Lewis bases.

BF ₃ _____	a. Lewis acid or Lewis base?
CCl ₄ _____	b. Lewis acid or Lewis base?
Al(CH ₃) ₃ _____	c. Is the element in position a a metallic element? Is it characterized by the chemical behavior of the metallic element?
- Classify the elements below as metals, metalloids, or nonmetals.

Rb _____	a. Is it a metallic element? Is it a metal?
Ag _____	b. Is it a metallic element? Is it a metal?
Te _____	c. The group number indicates it is a _____ element.
Bi _____	d. Is it a metallic element? Is it a metal?
Br _____	e. Is it a metallic element? Is it a metal?
Se _____	f. Is it a metallic element? Is it a metal?
Cl _____	g. Does it have metallic characteristics?
As _____	h. Does it have metallic characteristics?
Ge _____	i. Does it have metallic characteristics?
- Using only the periodic table, write the symbols for the elements in the following categories or periods.

He _____	a. Noble gas with the largest atomic mass.
Li _____	b. Noble gas with the largest atomic mass.
Ca _____	c. Noble gas with the largest atomic mass.
Transition _____	d. Noble gas with the largest atomic mass.
Al _____	e. Noble gas with the largest atomic mass.
Hydrogen _____	f. Is it a metallic or nonmetallic element? Is it a metal or nonmetal?

Chapter 5 Review The Periodic Law Mixed Answer Key

M Walker



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