

# 7.3

## BONDING IN METALS

### Section Review

#### Classify

- 1. Identify which atoms contribute to the metallic bond.
- 2. Describe the delocalized electron model.
- 3. Explain the bonding in metals.

#### Classify

- 1. Identify metals.
- 2. Classify.

#### Part A: Completion

Use the periodic table to determine which elements are metals. Classify each metal as a transition metal, main-group metal, or lanthanide.

- 1.   $\text{Ca}$  (main-group metal)
- 2.   $\text{Fe}$  (transition metal)
- 3.   $\text{Cu}$  (transition metal)
- 4.   $\text{Al}$  (main-group metal)
- 5.   $\text{Zn}$  (transition metal)
- 6.   $\text{Mg}$  (main-group metal)
- 7.   $\text{Ni}$  (transition metal)
- 8.   $\text{Ag}$  (transition metal)
- 9.   $\text{Au}$  (transition metal)
- 10.   $\text{Pt}$  (transition metal)
- 11.   $\text{Ba}$  (main-group metal)
- 12.   $\text{La}$  (lanthanide)
- 13.   $\text{Ce}$  (lanthanide)
- 14.   $\text{Pr}$  (lanthanide)
- 15.   $\text{Nd}$  (lanthanide)
- 16.   $\text{Pm}$  (lanthanide)
- 17.   $\text{Sm}$  (lanthanide)
- 18.   $\text{Eu}$  (lanthanide)
- 19.   $\text{Gd}$  (lanthanide)
- 20.   $\text{Dy}$  (lanthanide)
- 21.   $\text{Ho}$  (lanthanide)
- 22.   $\text{Er}$  (lanthanide)
- 23.   $\text{Tm}$  (lanthanide)
- 24.   $\text{Yb}$  (lanthanide)
- 25.   $\text{Lu}$  (lanthanide)

- 26.   $\text{Sc}$  (transition metal)
- 27.   $\text{Ti}$  (transition metal)
- 28.   $\text{V}$  (transition metal)
- 29.   $\text{Cr}$  (transition metal)
- 30.   $\text{Mn}$  (transition metal)
- 31.   $\text{Co}$  (transition metal)
- 32.   $\text{Ni}$  (transition metal)
- 33.   $\text{Cu}$  (transition metal)
- 34.   $\text{Zn}$  (transition metal)
- 35.   $\text{Ga}$  (main-group metal)
- 36.   $\text{In}$  (main-group metal)
- 37.   $\text{Tl}$  (main-group metal)
- 38.   $\text{Pb}$  (main-group metal)
- 39.   $\text{Bi}$  (main-group metal)
- 40.   $\text{Po}$  (main-group metal)
- 41.   $\text{At}$  (main-group metal)
- 42.   $\text{Rn}$  (noble gas)
- 43.   $\text{Ac}$  (actinide)
- 44.   $\text{Th}$  (actinide)
- 45.   $\text{Pa}$  (actinide)
- 46.   $\text{U}$  (actinide)
- 47.   $\text{Np}$  (actinide)
- 48.   $\text{Pu}$  (actinide)
- 49.   $\text{Am}$  (actinide)
- 50.   $\text{Cm}$  (actinide)
- 51.   $\text{Bk}$  (actinide)
- 52.   $\text{Cf}$  (actinide)
- 53.   $\text{Es}$  (actinide)
- 54.   $\text{Fm}$  (actinide)
- 55.   $\text{Md}$  (actinide)
- 56.   $\text{No}$  (actinide)
- 57.   $\text{Lr}$  (actinide)

#### Part B: True/False

Classify each statement as true or false. If false, explain why.

- 1.  Metals are good conductors of heat and electricity.
- 2.  Metals are malleable and ductile.
- 3.  Metals have high melting points.
- 4.  Metals are shiny.
- 5.  Metals are good conductors of heat and electricity.
- 6.  Metals are malleable and ductile.
- 7.  Metals have high melting points.
- 8.  Metals are shiny.

# Bonding In Metals Section Review Answers Key

**Jonathan Turk, Amos Turk**



**Bonding In Metals Section Review Answers Key:**

## Decoding **Bonding In Metals Section Review Answers Key**: Revealing the Captivating Potential of Verbal Expression

In an era characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its capability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Bonding In Metals Section Review Answers Key**," a mesmerizing literary creation penned by a celebrated wordsmith, readers embark on an enlightening odyssey, unraveling the intricate significance of language and its enduring effect on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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