

REPORT SHEET  
**Acid-Base Titration**

LAB  
**10**

**A. Concentration of Acetic Acid in Vinegar**

1. Brand Heinz Volume 5.00 mL (% on label) 5.0 %  
 2. Molarity (M) of NaOH \_\_\_\_\_ M

	Trial 1	Trial 2	Trial 3
3. Initial NaOH level in buret	0.1 mL	0.2 mL	0.1 mL
4. Final NaOH level in buret	39.6 mL	41.2 mL	40.1 mL
5. Volume (mL) of NaOH used			
6. Average volume (mL)			
7. Average volume in liters (L)			

Show all calculations for the following questions.

8. Moles of NaOH used in titration \_\_\_\_\_ mole NaOH
9. Moles of  $\text{HC}_2\text{H}_3\text{O}_2$  neutralized by NaOH \_\_\_\_\_ mole  $\text{HC}_2\text{H}_3\text{O}_2$
10. Molarity of  $\text{HC}_2\text{H}_3\text{O}_2$  \_\_\_\_\_ M  $\text{HC}_2\text{H}_3\text{O}_2$
11. Grams of  $\text{HC}_2\text{H}_3\text{O}_2$  \_\_\_\_\_ g  $\text{HC}_2\text{H}_3\text{O}_2$
12. Percent (m/v)  $\text{HC}_2\text{H}_3\text{O}_2$  in vinegar \_\_\_\_\_ %  $\text{HC}_2\text{H}_3\text{O}_2$

# Acid Base Titration Lab Chem Fax Answers

**Kenneth W. Whitten, Carl B.  
Bishop, Kenneth D. Gailey**



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