

CHEMISTRY VIVA QUESTIONS WITH ANSWERS CLASS XII

[1] VOLUMETRIC ANALYSIS

Question.1.What is a standard solution?

Answer. A solution whose strength is known is called a standard solution.

Question.2.What is a normal solution?

Answer. A solution containing one gram-equivalent mass of the solute per litre of the solution is called a normal solution.

Question.3.What is the equivalent mass of KMnO_4 when it acts as oxidizing agent in acidic medium?

Answer. KMnO_4 loses 5 electrons per molecule, when it acts as oxidizing agent in the presence of acids. Therefore, its equivalent mass is one-fifth of its molecular mass.
volumetric-analysis-viva-questions-with-answers-1

Question.4.Is sodium hydroxide a primary standard?

Answer. No.

Question.5.Are 'molality' and 'molarity' same?

Answer. No, molality of a solution is defined as the number of moles of solute present in 1000 grams of the solution whereas molarity tells us about the number of moles of the solute present per litre of the solution.

Question.6.What would be the normality of 0.10M KMnO_4 ?

Answer. It will be $0.1 \times 5 = 0.5 \text{ N}$.

Question.7.What volume of 10M HCl must be diluted with water to get 1L of 1M HCl?

Answer. 0.1 L.

Question.8.What is the basicity of H_2SO_4 ?

Answer. 2.

Question.9.What is the relationship between normality (N), molarity (M), molecular mass and equivalent mass?

Answer. Normality \times Eq. Mass = Molarity \times Mol. Mass.

Question.10.Why front door of the balance is closed during weighing?

Answer. Opening the front door causes vibrations in the pan due to operator's breath which leads to inaccurate results.

Chemistry Practical Viva Questions Answers For Class 12

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Table of Contents Chemistry Practical Viva Questions Answers For Class 12

1. Understanding the eBook Chemistry Practical Viva Questions Answers For Class 12
 - The Rise of Digital Reading Chemistry Practical Viva Questions Answers For Class 12
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Practical Viva Questions Answers For Class 12
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Practical Viva Questions Answers For Class 12
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Practical Viva Questions Answers For Class 12
 - Personalized Recommendations
 - Chemistry Practical Viva Questions Answers For Class 12 User Reviews and Ratings
 - Chemistry Practical Viva Questions Answers For Class 12 and Bestseller Lists
5. Accessing Chemistry Practical Viva Questions Answers For Class 12 Free and Paid eBooks
 - Chemistry Practical Viva Questions Answers For Class 12 Public Domain eBooks
 - Chemistry Practical Viva Questions Answers For Class 12 eBook Subscription Services
 - Chemistry Practical Viva Questions Answers For Class 12 Budget-Friendly Options
6. Navigating Chemistry Practical Viva Questions Answers For Class 12 eBook Formats

- ePub, PDF, MOBI, and More
 - Chemistry Practical Viva Questions Answers For Class 12 Compatibility with Devices
 - Chemistry Practical Viva Questions Answers For Class 12 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry Practical Viva Questions Answers For Class 12
 - Highlighting and Note-Taking Chemistry Practical Viva Questions Answers For Class 12
 - Interactive Elements Chemistry Practical Viva Questions Answers For Class 12
 8. Staying Engaged with Chemistry Practical Viva Questions Answers For Class 12
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Practical Viva Questions Answers For Class 12
 9. Balancing eBooks and Physical Books Chemistry Practical Viva Questions Answers For Class 12
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Practical Viva Questions Answers For Class 12
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Chemistry Practical Viva Questions Answers For Class 12
 - Setting Reading Goals Chemistry Practical Viva Questions Answers For Class 12
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Chemistry Practical Viva Questions Answers For Class 12
 - Fact-Checking eBook Content of Chemistry Practical Viva Questions Answers For Class 12
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
 14. Embracing eBook Trends
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 - Interactive and Gamified eBooks

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