

$n_1 \rightarrow$ Lyman (1) (UV Region)
 $n_2 \rightarrow$ Balmer (2) (visible)
 $n_3 \rightarrow$ 3 Paschen
 $n_4 \rightarrow$ 4 Brackett
 $n_5 \rightarrow$ 5 p-fund.

Time period = $\frac{2\pi r}{v}$ $\propto \frac{n^3}{Z^2}$
 Frequency = $\frac{1}{T}$ $\propto \frac{Z^2}{n^3}$

No. of spectral lines = $\frac{(n_2 - n_1)(n_2 - n_1 + 1)}{2}$

$\Delta E \propto \frac{1}{n^2}$

No. of spectral lines if $n=1 = \frac{n(n-1)}{2}$

No. of spectral lines.
 Lyman = $n-1$, Balmer = $n-2$
 Paschen = $n-3$, Brackett = $n-4$
 p-fund = $n-5$

Energy of $e^- \rightarrow n+1$

Photoelectric effect.
 $E_{\text{photon}} = W_0 + KE$
 $h\nu_p = h\nu_0 + KE$

Stopping potential $\Rightarrow E_{\text{photon}} - W_0 = qV$

de-Broglie's Eq.

$2\pi r = n\lambda$

No. of waves made in n^{th} orbit = n

$\lambda = \frac{h}{p} \Rightarrow \lambda = \frac{h}{\sqrt{2m \cdot KE}} \quad \lambda = \frac{h}{\sqrt{2m \cdot qV}} \quad \lambda_e = \sqrt{\frac{150}{V}} \text{ \AA} = \frac{12.3}{\sqrt{V}} \text{ \AA}$

$\Delta x \Delta p \geq \frac{h}{4\pi} \quad \Delta n \cdot m \Delta v \geq \frac{h}{4\pi} \quad \Delta \theta \Delta \phi \geq \frac{h}{4\pi}$

$\theta \rightarrow$ angular disp.
 $\phi \rightarrow$ momentum

No. of subshells in n^{th} shell or
 no. of orbitals l values from = 0 to $(n-1)$

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Table of Contents Chemistry Note For Iit Jee

1. Understanding the eBook Chemistry Note For Iit Jee
 - The Rise of Digital Reading Chemistry Note For Iit Jee
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Note For Iit Jee
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Note For Iit Jee
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Note For Iit Jee
 - Personalized Recommendations

- Chemistry Note For Iit Jee User Reviews and Ratings
- Chemistry Note For Iit Jee and Bestseller Lists
- 5. Accessing Chemistry Note For Iit Jee Free and Paid eBooks
 - Chemistry Note For Iit Jee Public Domain eBooks
 - Chemistry Note For Iit Jee eBook Subscription Services
 - Chemistry Note For Iit Jee Budget-Friendly Options
- 6. Navigating Chemistry Note For Iit Jee eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemistry Note For Iit Jee Compatibility with Devices
 - Chemistry Note For Iit Jee Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry Note For Iit Jee
 - Highlighting and Note-Taking Chemistry Note For Iit Jee
 - Interactive Elements Chemistry Note For Iit Jee
- 8. Staying Engaged with Chemistry Note For Iit Jee
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Note For Iit Jee
- 9. Balancing eBooks and Physical Books Chemistry Note For Iit Jee
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Note For Iit Jee
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chemistry Note For Iit Jee
 - Setting Reading Goals Chemistry Note For Iit Jee
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemistry Note For Iit Jee
 - Fact-Checking eBook Content of Chemistry Note For Iit Jee

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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