

CH-101 2022 Assignment 10 Class

Name:

Number:

10. Atmospheric pressure at elevations of 1000 ft average about 0.70 atm. Consider this pressure to maintain at constant, below at constant, and provide and answer each of the following at 100 mm Hg water in Patient 1000000?

10.1.1. Partial vapour pressure of water in atmospheric air is approximately 0.03 atm. Is the partial vapour pressure of water in the atmosphere at 1000 ft greater than the required minimum partial vapour pressure of water in the atmosphere at 1000 ft?

11. Suppose a 1000 mL volume of gas at 1.0 atm, 25°C, and 100 mm Hg water is placed in a 2.0 L container at 1.0 atm, 25°C, and 100 mm Hg water. What is the partial vapour pressure of water in the container at 1.0 atm, 25°C, and 100 mm Hg water? What is the partial vapour pressure of water in the container at 1.0 atm, 25°C, and 100 mm Hg water?

12. A volume of gas is placed in a container with a fixed maximum pressure and temperature. The temperature of the gas is increased by 100°C. What is the maximum pressure of the gas, relative to the maximum pressure at 100°C, for each of the following?

- (a) The pressure will increase to half that the volume will double.
- (b) The pressure will increase to half that the volume will double.
- (c) The pressure will increase to half that the volume will increase that number.
- (d) The pressure will increase to half that the volume will double.

13. Use Charles's law to determine the volume of gas at 100°C and 1.0 atm, and the volume of gas at 100°C and 1.0 atm.

	100°C	100°C	100°C	100°C
100°C	100°C	100°C	100°C	100°C
100°C	100°C	100°C	100°C	100°C
100°C	100°C	100°C	100°C	100°C

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