

## Chemistry Chapter 8 Test

### Multiple Choice

Identify the choice that best completes the statement or answers the question.

- \_\_\_\_ 1. What type of bond would you expect in a compound of oxygen and sodium?  
 a. ionic  
 b. metallic  
 c. covalent  
 d. polar covalent
- \_\_\_\_ 2. The symbols  $\delta^-$  and  $\delta^+$  indicate which of the following?  
 a. electronegativity  
 b. electrostatic force  
 c. charge  
 d. partial charge
- \_\_\_\_ 3. Which of the following represents a nonpolar covalent molecule?  
 a. NaCl  
 b. Br<sub>2</sub>  
 c. HCN  
 d. Au
- \_\_\_\_ 4. The measure of an atom's ability to attract electrons to itself in a chemical bond is called \_\_\_\_\_.  
 a. electron affinity  
 b. polarity  
 c. electronegativity  
 d. ionization energy
- \_\_\_\_ 5. The polarity of a bond depends on the difference in the atoms' \_\_\_\_\_.  
 a. electron affinities  
 b. electronegativities  
 c. atomic radii  
 d. ionization energies
- \_\_\_\_ 6. What type of bond would you expect to form between nitrogen and potassium?  
 a. ionic  
 b. covalent  
 c. polar covalent  
 d. metallic
- \_\_\_\_ \* High electronegativity differences are found in \_\_\_\_\_ bonds.  
 a. metallic  
 b. ionic metal - nonmetal  
 c. covalent  
 d. Pauling
- \_\_\_\_ \* To achieve an octet when combining with potassium, would an atom of a Group 16 element be expected to lose, gain, or share electrons?  
 a. lose  
 b. gain  
 c. share  
 d. This cannot be determined from the given information.
- \_\_\_\_ 9. Which of the following does not exist as a diatomic molecule?  
 a. argon  
 b. hydrogen  
 c. chlorine  
 d. oxygen  
 H N O F Cl Br I  
 Gas: H<sub>2</sub> N<sub>2</sub> O<sub>2</sub> F<sub>2</sub> Cl<sub>2</sub> Br<sub>2</sub> I<sub>2</sub>
- \_\_\_\_ 10. Which of these is not a covalent molecule?  
 a. F<sub>2</sub>  
 b. CO  
 c. LiF  
 d. NH<sub>3</sub>
- \_\_\_\_ 11. A dash in a Lewis structure represents \_\_\_\_\_.  
 a. one electron  
 b. two electrons  
 c. four electrons  
 d. unpaired electrons
- \_\_\_\_ 12. The central atom in a Lewis structure is usually the atom \_\_\_\_\_.

# Chapter 5 Chemistry Test Answers

**Clifford Lane**



## **Chapter 5 Chemistry Test Answers:**

## Chapter 5 Chemistry Test Answers Book Review: Unveiling the Power of Words

In a global driven by information and connectivity, the power of words has be much more evident than ever. They have the ability to inspire, provoke, and ignite change. Such could be the essence of the book **Chapter 5 Chemistry Test Answers**, a literary masterpiece that delves deep in to the significance of words and their affect our lives. Published by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we will explore the book is key themes, examine its writing style, and analyze its overall affect readers.

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