

## Charge Calculation Exercise

A foundry needs to produce a heat of 2 Tons of 6061 alloy. They have prepared a melt with 80% of total required tonnage using raw materials in foundry. Composition of this melt is given in following table:

Element	Mg	Si	Iron	Cu	Zn	Ti	Mn	Cr	Al
Wt.%	0.5	0.3	0.5	0.1	0.2	0.1	0.1	0.2	rest

1. Search for required elemental composition (wt.%) of 6061.
2. Then Calculate the required amounts of master alloys that need to be added to obtain the required wt.%. Present data in tabular form.
3. You should also comment on addition method of these master alloys after reading some literature.
4. Three bonus marks are for neatness and proper management of the assignment.

**Note:** Available master alloys of Aluminum in this foundry are:

Mg15%; Cr5%; Mn10%; Si20%; Fe2%; Cu5%

# Foundry Charge Calculation

**John Brown**



## **Foundry Charge Calculation:**

## The Enigmatic Realm of **Foundry Charge Calculation**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing short of extraordinary. Within the captivating pages of **Foundry Charge Calculation** a literary masterpiece penned with a renowned author, readers attempt a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting effect on the hearts and minds of those that partake in its reading experience.

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