

Semester 1 Exam Study Guide KEY

2. The following types of Chemistry mainly study:
- inorganic chemistry: chemicals that do not contain carbon
 - analytical chemistry: the composition/make-up of substances
 - biochemistry: the relationship between biology and chemistry (ex. Movement of athletes)
 - organic chemistry: chemicals that contain carbon
 - physical chemistry: rate of chemical reactions

3. Scientific Method

a. Steps of the Scientific Method

- Question/Problem
- Hypothesis
- Experiment
- Analyze Data
- Conclusion

b. Types of Variables

- Dependent variable: on the y-axis-measured
- Independent variable: on the x-axis- manipulating
- Control: the normal "situation" where nothing is changed
- Constants are conditions that are held constant throughout the experiment

c. Accuracy vs. Precision

- Precision: multiple measurements are close together to one another
- Accuracy: measurements are close to the actual value

d. Percent error

- Percent error equation = $\frac{|\text{obs. value} - \text{accepted value}|}{\text{accepted}}$ x 100%

3. Lab Equipment and Safety

a. *Please see lab equipment sheet for pictures and uses

- Buret
- Graduated cylinder
- Erlenmeyer flask
- Beaker
- Crucible

b. Lab Safety basic rules

- One example, "Do not put laughing gas in water"

4. Matter and Measurements

a. Separation of Mixtures

- Filtration: separates solid from a liquid based on physical properties
- Distillation: separates a liquid from a liquid based on boiling points

b. Types of Mixtures

- Homogeneous mixture has one phase (level) of its components (solution)

Midterm Fall Semester Exam Study Guide

Rachel Sandford



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