

**Introduction**

This exam is a closed book examination. You are permitted to use a calculator. However, you are not permitted to use any other aids such as notes, textbooks, or any other electronic devices. The exam is divided into two sections: Section A (Multiple Choice) and Section B (Short Answer). The total time for the exam is 1 hour. You are permitted to use a calculator for the entire duration of the exam. The questions are of varying difficulty and are designed to test your understanding of the course material.

**Instructions:** Read the questions carefully and answer them in the spaces provided. Write your answers in the spaces provided. Do not write on the back of the page. You are permitted to use a calculator for the entire duration of the exam.

**Part A: Multiple Choice Questions**



1. Which of the following is a physical property of a substance?
  - a. Flammability
  - b. Boiling point
  - c. Reactivity
  - d. Solubility
2. The chemical formula for water is  $H_2O$ . How many oxygen atoms are present in one molecule of water?
  - a. 1
  - b. 2
  - c. 3
  - d. 4
3. The chemical formula for sodium chloride is  $NaCl$ . How many sodium atoms are present in one molecule of sodium chloride?
  - a. 1
  - b. 2
  - c. 3
  - d. 4



4. Which of the following is a chemical property of a substance?
  - a. Melting point
  - b. Boiling point
  - c. Flammability
  - d. Solubility
5. The chemical formula for carbon dioxide is  $CO_2$ . How many carbon atoms are present in one molecule of carbon dioxide?
  - a. 1
  - b. 2
  - c. 3
  - d. 4

Question	Answer A	Answer B	Answer C	Answer D	Answer E
1.	Flammability	Boiling point	Reactivity	Solubility	Boiling point
2.	1	2	3	4	Boiling point
3.	1	2	3	4	Boiling point
4.	Flammability	Boiling point	Reactivity	Solubility	Boiling point
5.	1	2	3	4	Boiling point

6. The chemical formula for sodium chloride is  $NaCl$ . How many sodium atoms are present in one molecule of sodium chloride?
  - a. 1
  - b. 2
  - c. 3
  - d. 4
7. The chemical formula for carbon dioxide is  $CO_2$ . How many carbon atoms are present in one molecule of carbon dioxide?
  - a. 1
  - b. 2
  - c. 3
  - d. 4

**Part B: Short Answer Questions**



8. Write the chemical formula for water.
9. Write the chemical formula for carbon dioxide.



10. Complete the table below with the missing values for the given chemical reactions.

Reaction	Reactants	Products	State	Balance
1.	$H_2$	$O_2$	Gas	Not balanced
2.	$NaCl$	$Na$	Solid	Not balanced
3.	$CO_2$	$C$	Solid	Not balanced
4.	$H_2O$	$H_2$	Gas	Not balanced

Write the chemical formula for water:  $H_2O$

# Basic Stoichiometry Phet Lab Homework Exercises Answers

**J Rink**



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