

**Guided Practice - Acids & Bases****Properties of Acids and Bases**

1. Match the following terms with their descriptions:

- |                        |       |   |
|------------------------|-------|---|
| a. Arrhenius Acid      | _____ | the substance formed when a proton is added to _____        |
| b. Arrhenius Base      | _____ | the remaining substance when a proton is lost _____         |
| c. Conjugate Acid      | _____ | from an acid _____  |
| d. Conjugate Base      | _____ | a substance that produces OH <sup>-</sup> in solution _____ |
| e. Bronsted-Lowry Acid | _____ | a substance producing H <sup>+</sup> in solution _____      |
| f. Bronsted-Lowry Base | _____ | a proton acceptor _____                                     |
|                        |       | a proton donor _____  |

2. Which of the following represent conjugate acid-base pairs? Label each species as a conjugate acid or a conjugate base. For those pairs that are NOT conjugates, write the correct conjugate acid or base for each species in the pair, labeling each species correctly.

- $\text{HSO}_4^-$ ;  $\text{SO}_4^{2-}$
- $\text{HBr}$ ;  $\text{BrO}^-$
- $\text{H}_3\text{PO}_4$ ;  $\text{PO}_4^{3-}$
- $\text{H}_2\text{O}$ ;  $\text{HO}^-$

3. Complete the following table regarding acid strength.

Property	Strong Acid	Weak Acid
Definition		
Describe Dissociation		
Species present in solution		
Relative strength of conjugate base compared to water		
Equation showing acid's reaction to water		

# Acids Bases And Solutions Answer Key Lab35

**František Vydra, Karel Štulík, Eva  
Juláková**

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Keyed to the learning goals in the text this guide is designed to promote active learning through a variety of exercises with answers and mastery exams The guide also contains complete solutions to odd numbered problems Experiments to Study Our Atmospheric Environment Steven Businger, 1996 Consisting mainly of hands on experiments this work is designed specifically to give students of diverse academic backgrounds an opportunity to explore and understand the underlying physical principles of meteorology *Nuclear Science Abstracts*, 1969-11 **Electroanalytical Abstracts**, 1970 International journal dealing with the documentation of all aspects of fundamental physico chemical and analytical electrochemistry Industrial Laboratory, 1976 **Journal of Analytical Chemistry of the USSR.**, 1973

**Electrochemical Stripping Analysis** František Vydra, Karel Štulík, Eva Juláková, 1976 Tsennaaks Stadii, bonjaaks slangii Heikki Paunonen, 1970 *Bulletin of Thermodynamics and Thermochemistry*, 1970 **Nachrichten aus Chemie, Technik und Laboratorium**, 1987 **Science Citation Index**, 1994 Vols for 1964 have guides and journal lists

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