

Cobalt(II) chloride dihydrate  
( $\text{CoCl}_2 \cdot 2\text{H}_2\text{O}$ )



(a) An amount of solute is weighed out on an analytical balance and then

(b) A portion of the solvent is added to the volumetric flask.

(c) The mixture is swirled until all of the solute is dissolved.

(d) Additional solvent is added up to the mark on the volumetric flask.

# Stock Solution Preparation

**Clifford Lane**



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