

Problems #1 thru #2: A researcher is testing the following hypotheses with $n = 1000$ and $\alpha = 0.01$:

$$H_0: \pi = 0.29$$

$$H_A: \pi \neq 0.29$$

1. What is the level of significance?

- A. 0.001
- B. 0.01
- C. 0.02
- D. 0.05
- E. 0.1

2. If the test statistic were 1.238, what is the p-value?

- A. 0.1003
- B. 0.1075
- C. 0.1093
- D. 0.2150
- E. 0.2186

$$.5 - .3925$$



Problems #3 thru #5: The results from a sample of net weights of 6 bottles filled by a machine manufactured by Edne, and a sample of the net weights of 8 bottles filled by a similar machine manufactured by Orno, Inc. are (in grams):

Company	Sample Size	Mean	Standard Deviation
Edne (E)	6	7	1.4
Orno, Inc. (O)	8	10	2.3

Use a 0.01 level of significance to test whether the mean net weight of bottles filled by the Edne machine is different than the mean net weight of bottles filled by the Orno, Inc. machine. Assume normality if needed.

3. What are the correct hypotheses?

- A. $H_0: \mu_E - \mu_O \neq 0$ $H_A: \mu_E - \mu_O = 0$
- B. $H_0: \mu_E - \mu_O > 0$ $H_A: \mu_E - \mu_O \leq 0$
- C. $H_0: \mu_E - \mu_O \leq 0$ $H_A: \mu_E - \mu_O > 0$
- D. $H_0: \mu_E - \mu_O \geq 0$ $H_A: \mu_E - \mu_O < 0$
- E. $H_0: \mu_E - \mu_O = 0$ $H_A: \mu_E - \mu_O \neq 0$

4. What are the degrees of freedom?

- A. ∞
- B. 14
- C. 13
- D. 12
- E. 11

5. What is the pooled variance, s_p^2 ?

- A. 4.5067
- B. 3.9025
- C. 2.2333
- D. 1.9755
- E. 1.9250

$$\frac{(6-1)(1.4)^2 + (8-1)(2.3)^2}{12}$$

$$9.8 + 37.03 = \frac{46.83}{12}$$

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web hydrogen will be left slightly positive so this is a polar molecule i represent this as a couple of different animals here got a bear and it s got the five valence electrons and it shares the single electrons with its penguin friend the penguins each have one valence electron to be shared with from the bear

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web answer because the two polar bears or penguins have equal electronegativity they will pull electrons equally hard and in most cases neither will end up with more electrons each polar bear and penguin has an electronegativity of the same value a non polar bond is formed when two polar bears or two penguins are bonded together

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web to answer this question you will examine 1 electronegativity 2 nonpolar vs polar molecules 3 electronegativity and bonding electronegativity the h atom and the cl atom in hydrogen chloride

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web 1 how does the comic strip define a polar molecule 2 define electronegativity as you understand it after reading the first two pages of the comic strip 3 what is the artist trying to represent by two polar bears arm wrestling or two penguins arm wrestling 4 what three types of bonds are represented on the third page of the comic strip

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web so the final representation of the NH_3 molecule using polar bears and penguins is as follows begin array c text polar bear nitrogen text with two lone pairs of electrons text and three penguins hydrogen end array this is a fun and creative way to represent the NH_3 molecule

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web polar bears and penguins purpose in this lesson you will be exploring polarity and bonding between atoms in greater detail a comic book will provide new information about these topics and will introduce you to the concept of electronegativity which helps us understand partial charges