

Radioactive Decay Statistics

In today's lab we are simulating the decay of a radioactive isotope of an element into a more stable form. This is represented by one color of candy (M&M) labeled B undergoing transmutation (trans = across, mutation = change) into another color of candy labeled A. During this process we monitor the percent of the color B remaining in the beaker and use a graph to find the Half-Life of this color in the beaker.

Procedure:

1. Decide which colors are type A and type B for your experiment. Then count out 50 of each and mix into one beaker. Count out another 100 pieces of type A into a separate beaker to use as replacements.
2. Perform Run 1: Blindly remove one piece from the A & B mixture. Replace it with a Type A, no matter what color came out. Mix the candy gently. Repeat until 10 pieces had been taken out. Record results from this Run in the data table below.
3. Repeat step 2 for Runs 2 through 10. In each Run, 10 pieces are removed from the mixture of colors and always replaced with a type A. Record the results for each run in the data table below.
4. Label numbers 1 through 10 Runs on the x-axis of the semi-log graph paper. Label the percentage of Type B Remaining on the y-axis of the graph paper.
 - a. Note: y-axis starts at 1% and is not linear.
5. Plot % Type B Remaining for each Run.
6. Draw the best straight line through the points. From the graph, read the time or Run value where the straight line crosses the 50% line. This is the Type B Half-Life.

Type B Half-Life: 6.5

Data: fill this out for the data given and graph it on the next page

Run	# Type A removed	# Type B Removed	# Type B Remaining	% Type B Remaining
1	6	4	46	92
2	5	5	41	82
3	7	3	38	76
4	6	4	34	68
5	5	5	29	58
6	7	3	26	52
7	7	3	23	46
8	8	2	21	42
9	9	1	20	40
10	9	1	19	38

Modeling Radioactive Decay Lab Answers

**Judy Flagg Moran, Marsha Jane
Davis, Mary E. Murphy**



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Table of Contents Modeling Radioactive Decay Lab Answers

1. Understanding the eBook Modeling Radioactive Decay Lab Answers
 - The Rise of Digital Reading Modeling Radioactive Decay Lab Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Modeling Radioactive Decay Lab Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Modeling Radioactive Decay Lab Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Modeling Radioactive Decay Lab Answers
 - Personalized Recommendations
 - Modeling Radioactive Decay Lab Answers User Reviews and Ratings
 - Modeling Radioactive Decay Lab Answers and Bestseller Lists
5. Accessing Modeling Radioactive Decay Lab Answers Free and Paid eBooks
 - Modeling Radioactive Decay Lab Answers Public Domain eBooks
 - Modeling Radioactive Decay Lab Answers eBook Subscription Services
 - Modeling Radioactive Decay Lab Answers Budget-Friendly Options

6. Navigating Modeling Radioactive Decay Lab Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Modeling Radioactive Decay Lab Answers Compatibility with Devices
 - Modeling Radioactive Decay Lab Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Modeling Radioactive Decay Lab Answers
 - Highlighting and Note-Taking Modeling Radioactive Decay Lab Answers
 - Interactive Elements Modeling Radioactive Decay Lab Answers
8. Staying Engaged with Modeling Radioactive Decay Lab Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Modeling Radioactive Decay Lab Answers
9. Balancing eBooks and Physical Books Modeling Radioactive Decay Lab Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Modeling Radioactive Decay Lab Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Modeling Radioactive Decay Lab Answers
 - Setting Reading Goals Modeling Radioactive Decay Lab Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Modeling Radioactive Decay Lab Answers
 - Fact-Checking eBook Content of Modeling Radioactive Decay Lab Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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