

NAME: _____

SECTION: _____

TIME: _____

Matching: Answer each question correctly. Place the letter that best matches the description by writing the letter in the corresponding blank. (10 points each)

- | | | | |
|-----------|---|---------------------------------|--------------------------|
| <u>1</u> | A | A type of covalent bond that is | A. Ionic |
| <u>2</u> | B | A type of ionic bond that is | B. Dipole-dipole |
| <u>3</u> | C | A type of ionic bond that is | C. Dipole-induced dipole |
| <u>4</u> | D | A type of ionic bond that is | D. Hydrogen bond |
| <u>5</u> | E | A type of ionic bond that is | E. Dipole-dipole |
| <u>6</u> | F | A type of ionic bond that is | F. Dipole-induced dipole |
| <u>7</u> | G | A type of ionic bond that is | G. Hydrogen bond |
| <u>8</u> | H | A type of ionic bond that is | H. Dipole-dipole |
| <u>9</u> | I | A type of ionic bond that is | I. Dipole-induced dipole |
| <u>10</u> | J | A type of ionic bond that is | J. Hydrogen bond |

Multiple Choice: Each question asks you to identify the correct choice. (1 point each)

- 11 A. The number of lone pairs on the central atom in the following molecule is
- a. 0
 - b. 1
 - c. 2
 - d. 3
 - e. 4
- 12 B. The number of lone pairs on the central atom in the following molecule is
- a. 0
 - b. 1
 - c. 2
 - d. 3
 - e. 4

Assessment 4 2 Module 4 Investing Answers

RS Peters



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