

Acids, Bases, Solutions Study Guide

4 pages + 4-page Answer Key

Chemistry

Physical Science

SOLUTIONS, ACIDS, AND BASES STUDY GUIDE

What is a homogeneous mixture? Provide two examples.

Describe the characteristics of a saturated solution.

1. Define the terms solvent and solute as they apply to solutions.

2. Define solubility, and how is it measured?

3. Distinguish between concentrated and dilute.

4. List the properties of bases and acids.

5. Explain the significance of the pH scale.

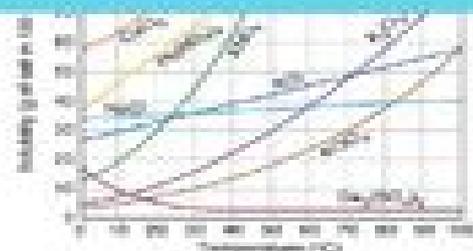
6. How does the concentration of an acid affect its pH?

7. Explain the relationship between pH and pOH.

8. Explain the relationship between pH and pKa.

9. Explain the relationship between pH and pKb.

10. Explain the relationship between pH and pKsp.



11. Describe how you would use the graph to determine the amount of Na_2CO_3 that can be dissolved in 100 grams of water at 40°C.

12. Which salt has the greatest solubility at 40°C?



ANSWER KEY INCLUDED

Acids And Bases Study Guide Key

G Psacharopoulos



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