



### Step 6 & Final step

#### Chloride Content Calculation in Water

- Weight of sample ( $V_2$ ) = 50 ml
- Titre value of water sample ( $V_1$ ) = 19.3
- Titre value of blank ( $V_3$ ) = 0.5

$$\text{Chloride, mg/l} = \frac{(V_1 - V_3) \times N \times 35.45 \times 1000}{V_2}$$
$$= \frac{(19.3 - 0.5) \times 0.01 \times 35.45 \times 1000}{50}$$
$$= 127.94 \text{ mg/l}$$

# Determination Of Chloride Using Potentiometry Asdl Home

**Thomas Griffiths**



## **Determination Of Chloride Using Potentiometry Asdl Home:**

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