

## Chapter 25 Nuclear Chemistry Practice Test

### Multiple Choice

Identify the choice that best completes the statement or answers the question.

- \_\_\_\_ 1. What does the 4 in  ${}^4_2\text{He}$  represent?  
a. the mass number  
b. the atomic number  
c. the number of protons  
d. the number of neutrons
- \_\_\_\_ 2. What does the 101 in  ${}^{256}_{101}\text{Md}$  represent?  
a. the mass number  
b. the atomic number  
c. the nuclide number  
d. the number of neutrons
- \_\_\_\_ 3. Balance the following equation:  ${}^{226}_{88}\text{Rn} \rightarrow {}^{226}_{86}\text{Rn} + \text{_____}$   
a.  ${}^4_2\text{He}$   
b.  ${}^1_0e^-$   
c.  ${}^1_1\text{H}$   
d.  ${}^0_{-1}e^-$
- \_\_\_\_ 4. Balance the following equation:  ${}^{239}_{93}\text{Np} \rightarrow \text{_____} + {}^0_{-1}e^-$   
a.  ${}^{239}_{90}\text{Th}$   
b.  ${}^{239}_{92}\text{U}$   
c.  ${}^{239}_{94}\text{Pu}$   
d.  ${}^{238}_{94}\text{Pu}$
- \_\_\_\_ 5. The spontaneous disintegration of a nucleus into a slightly lighter and more stable nucleus, accompanied by emission of particles, electromagnetic radiation, or both, is  
a. nuclear fusion.  
b. nuclear radiation.  
c. radioactive decay.  
d. nuclear fission.
- \_\_\_\_ 6. Which of the following forms of radiation has the greatest penetrating power?  
a. alpha particles  
b. beta particles  
c. gamma rays  
d. positrons
- \_\_\_\_ 7. Alpha particles are  
a. electrons.  
b. helium nuclei.  
c. electromagnetic waves.  
d. neutrons.
- \_\_\_\_ 8. Beta particles are  
a. electrons.  
b. helium nuclei.  
c. electromagnetic waves.  
d. neutrons.
- \_\_\_\_ 9. Which of the illustrations below represents a fission reaction?

# Chapter 25 Nuclear Chemistry Practice Problems

## Answer Key

**Institute of Marine Engineers**



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## **Table of Contents Chapter 25 Nuclear Chemistry Practice Problems Answer Key**

1. Understanding the eBook Chapter 25 Nuclear Chemistry Practice Problems Answer Key
  - The Rise of Digital Reading Chapter 25 Nuclear Chemistry Practice Problems Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Chapter 25 Nuclear Chemistry Practice Problems Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chapter 25 Nuclear Chemistry Practice Problems Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chapter 25 Nuclear Chemistry Practice Problems Answer Key
  - Personalized Recommendations
  - Chapter 25 Nuclear Chemistry Practice Problems Answer Key User Reviews and Ratings

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- Chapter 25 Nuclear Chemistry Practice Problems Answer Key and Bestseller Lists
- 5. Accessing Chapter 25 Nuclear Chemistry Practice Problems Answer Key Free and Paid eBooks
  - Chapter 25 Nuclear Chemistry Practice Problems Answer Key Public Domain eBooks
  - Chapter 25 Nuclear Chemistry Practice Problems Answer Key eBook Subscription Services
  - Chapter 25 Nuclear Chemistry Practice Problems Answer Key Budget-Friendly Options
- 6. Navigating Chapter 25 Nuclear Chemistry Practice Problems Answer Key eBook Formats
  - ePub, PDF, MOBI, and More
  - Chapter 25 Nuclear Chemistry Practice Problems Answer Key Compatibility with Devices
  - Chapter 25 Nuclear Chemistry Practice Problems Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Chapter 25 Nuclear Chemistry Practice Problems Answer Key
  - Highlighting and Note-Taking Chapter 25 Nuclear Chemistry Practice Problems Answer Key
  - Interactive Elements Chapter 25 Nuclear Chemistry Practice Problems Answer Key
- 8. Staying Engaged with Chapter 25 Nuclear Chemistry Practice Problems Answer Key
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Chapter 25 Nuclear Chemistry Practice Problems Answer Key
- 9. Balancing eBooks and Physical Books Chapter 25 Nuclear Chemistry Practice Problems Answer Key
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Chapter 25 Nuclear Chemistry Practice Problems Answer Key
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Chapter 25 Nuclear Chemistry Practice Problems Answer Key
  - Setting Reading Goals Chapter 25 Nuclear Chemistry Practice Problems Answer Key
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chapter 25 Nuclear Chemistry Practice Problems Answer Key
  - Fact-Checking eBook Content of Chapter 25 Nuclear Chemistry Practice Problems Answer Key
  - Distinguishing Credible Sources

13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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