

Chemistry of Life

DOODLE NOTES

chemistry of life
BASIC CHEMISTRY REVIEW

atoms and life
All living things are made of atoms.
An atom is the smallest part of an element. Many different types of atoms exist and they are arranged in the periodic table of elements. Atoms are made up of subatomic particles.

Label the structure of this oxygen atom:

• Note the charge and mass of each subatomic particle.

Charge	Mass	Electron	Proton	Neutron
-1	1/1836	○	●	●

• How many shells are in the first shell? 2
• How many shells are in the outer shell? 2
• What do we call electrons in the outer shell? Valence electrons

elements
An element is a substance that is made of one type of atom only. They cannot be broken down into simpler substances.

compounds
A compound is made of two or more types of atoms that are chemically bonded. They can be broken down chemically. There are two categories: ionic and covalent.

mixtures
A mixture is a mix of two or more types of atoms that are NOT chemically bonded. They can be separated physically. There are two categories: homogeneous and heterogeneous.

Examples:
Helium - He
Sulfur - S
Sodium Chloride - NaCl
Oxide - O₂

ionic
An ionic bond is formed from a metal and a non-metal. It is made of oppositely charged ions. The ions are held together by electrostatic forces.

Examples:
Sodium Chloride (NaCl)
Oxide (O²⁻)

covalent
A covalent bond is formed from two non-metals. It is made of atoms that share their outer shell electrons. The atoms are held together by the shared pair of electrons.

Examples:
Water (H₂O)
Methane (CH₄)

functional groups
These are groups of atoms that are attached to a carbon atom. They give the molecule its characteristic properties.

Hydroxyl group
-OH

Carbonyl group
C=O

Carboxyl group
-COOH

Amino group
-NH₂

Phosphate group
-PO₄

Acidic group
-COOH



MULTIPLE VERSIONS & EXAMPLE

Biology The Chemistry Of Life Answer Key

Vera M. Kolb



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Table of Contents Biology The Chemistry Of Life Answer Key

1. Understanding the eBook Biology The Chemistry Of Life Answer Key
 - The Rise of Digital Reading Biology The Chemistry Of Life Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Biology The Chemistry Of Life Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Biology The Chemistry Of Life Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Biology The Chemistry Of Life Answer Key
 - Personalized Recommendations
 - Biology The Chemistry Of Life Answer Key User Reviews and Ratings
 - Biology The Chemistry Of Life Answer Key and Bestseller Lists
5. Accessing Biology The Chemistry Of Life Answer Key Free and Paid eBooks

- Biology The Chemistry Of Life Answer Key Public Domain eBooks
 - Biology The Chemistry Of Life Answer Key eBook Subscription Services
 - Biology The Chemistry Of Life Answer Key Budget-Friendly Options
6. Navigating Biology The Chemistry Of Life Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Biology The Chemistry Of Life Answer Key Compatibility with Devices
 - Biology The Chemistry Of Life Answer Key Enhanced eBook Features
 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Biology The Chemistry Of Life Answer Key
 - Highlighting and Note-Taking Biology The Chemistry Of Life Answer Key
 - Interactive Elements Biology The Chemistry Of Life Answer Key
 8. Staying Engaged with Biology The Chemistry Of Life Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Biology The Chemistry Of Life Answer Key
 9. Balancing eBooks and Physical Books Biology The Chemistry Of Life Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Biology The Chemistry Of Life Answer Key
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Biology The Chemistry Of Life Answer Key
 - Setting Reading Goals Biology The Chemistry Of Life Answer Key
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Biology The Chemistry Of Life Answer Key
 - Fact-Checking eBook Content of Biology The Chemistry Of Life Answer Key
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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