

## Question bank biology unit test biology collage

### Q.1 Multiple Choice Questions

1. Which of following can used to explain the law of independent assortment .....

  - A) Monohybrid Cross
  - B) Test Cross
  - C) Dihybrid Cross
  - D) Back Cross

**C) Dihybrid Cross**

2. The Process of development of Zygote into an embryo is called .....

  - A) Embryogenesis
  - B) Karyokinesis
  - C) Sporogenesis
  - D) Parthenogenesis

**A) Embryogenesis**

3. \_\_\_\_\_ is the Convenient and Cheap method of Artificial Vegetative Propagation .

  - A) Grafting
  - B) Budding
  - C) Cutting
  - D) Micropropagation

**C) Cutting**

4. If the organism Show dominant trait of a Parent but are genetically different they are

  - A) Heterozygous
  - B) Homozygous
  - C) Hemizygous
  - D) Heterologous

**A) Heterozygous**

5. The gene which codes for a complete functional polypeptide that fully express itself it is called as \_\_\_\_\_

  - A) Recessive gene

# Biology Question Bank

**Oswaal Editorial Board**



**Biology Question Bank:**

## Reviewing **Biology Question Bank**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is really astonishing. Within the pages of "**Biology Question Bank**," an enthralling opus penned by a highly acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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being reduced will have electrons as reactants and the oxidized substance will have electrons as products usually all reactions are written as reduction reactions in half reaction tables

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web bonus problem here is a balanced net ionic equation 3br<sub>2</sub> 6oh<sup>-</sup> 5br<sup>-</sup> bro<sup>-</sup> 3h<sub>2</sub>o the above equation takes place in two stages first bromine reacts with oh ions to form bromide ions and bro ions second the bro ions react to form bromide ions and bromate v ions

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web we can balance it for charge by adding two electrons to the right side of the equation so that the net charge on each side is 0 oxidation ni s ni a 2 a q 2 e now that the oxidation half reaction is balanced it tells us that two electrons are produced for every atom of nickel oxidized but where do those electrons go

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reactions  $6e$   $14h$   $Cr_2O_7^{2-}$   $2Cr^{3+}$   $7H_2O$   $Fe^{2+}$   $Fe^{3+}$   $e^-$  2 equalize the electrons  $6e$   $14h$   $Cr_2O_7^{2-}$   $2Cr^{3+}$   $7H_2O$   $6Fe^{2+}$   $6Fe^{3+}$   $6e^-$  multiplied by a factor of 6

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