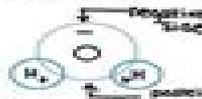


Structure of H₂O
hydrogen bonding
Consists of 1:8:1



unit one guide

Biological molecules
proteins
nucleic acids

H₂O provides cohesion, adhesion, surface tension, high specific heat, high heat of vaporization, excellent solvent, and buffer capacity.

hydrogen bonds require lots of energy to change state of matter

CONDENSATION - when two molecules are produced hydrogen and hydroxyl ions, H₂O leaves, pH - more hydrogen

OH⁻ attracts to some
OH⁻ attracted to other
OH⁻ hydrogen bonds
OH⁻ hydrogen bonds
OH⁻ hydrogen bonds
OH⁻ hydrogen bonds

C	H	N	O	P	S
Carbon	Hydrogen	Nitrogen	Oxygen	Phosphorus	Sulfur

more hydrogen
C - build biological molecules, cell formation
N - build proteins, nucleic acids
P - build nucleic acids + lipids
make up 98% of organisms

Carbohydrates ↓ C, H, O	Lipids ↓ C, H, O	Nucleic acids ↓ C, H, N, O, P	Proteins ↓ C, H, N, O
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Condensation synthesis is the removal of an H₂O molecule to bond monomers. Hydrolysis is the addition of a water molecule to break a bond between monomers (breaks protein).

Nucleic Acid Structure:

* the carbon sugar (deoxyribose/ribose), phosphate, nitrogen base (G, C, T, A), exact shape of strand



Starch vs Cellulose

Starch - alpha glucose
Cellulose - beta glucose

Proteins: specific order of amino acids
Primary structure

1st group: substructure determines structure + function



PROTEINS

amino acids bond via a peptide bond (covalent) through dehydration synthesis
1st group change can cause a hydrophobic protein, forces the place of a hydrophilic protein

membranes

* made up of phospholipids
* cross selectively permeable barrier
* contain phospholipids, cholesterol, and proteins



LIPIDS

Phospholipid macromolecules
- differences in saturation determine structure and function
- Cells, steroids, phospholipids
- Saturated - no double bonds
- Unsaturated - has double bonds
- 4 in 1: triglycerides, cholesterol, most of the membrane, steroids



Amino Acids: have directionality

* linear chains, connected by covalent bonds
* amino end and carboxyl end

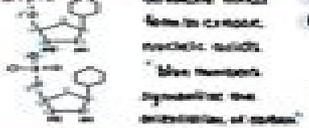
PROTEIN STRUCTURE

Primary: sequence of amino acids, specific order determines protein shape
Secondary: how folding of chain into alpha helix and beta sheet occurs
Tertiary: 3-D shape, hydrogen bonds, disulfide bond, ionic bonds, hydrophobic int.
Quaternary: interactions between multiple polypeptide chains

NUCLEIC ACIDS

* add new nucleotides to the 3' end
* two nitrogenous bases: purine + pyrimidine
* pyrimidine is single ringed, purine is double
* antiparallel double helix
* 5' on phosphate end, 3' on hydroxyl end
* both back and back have nitrogenous bases (phosphodiester) + sugar-phosphate backbone

DEHYDRATION SYNTHESIS



Chaperones:

* protein molecules that assist protein folding

ANTIPARALLEL



NH₂ = Amino
COOH = Carboxyl

ALDOSE VS KETOSE

* contains aldehyde functional group

ALPHA VS BETA

* contains ketone functional group



* all monomers are in the same organization



* Every glucose monomer is upside down from its neighbor

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