

Dilutions – Practice Problems

Instructions: Use the equation for dilutions to complete the problems below.

- 1) A student obtains 0.30 liters of 2.0 M silver nitrate solution. She dilutes the solution until it has a volume of 2.6 liters. What is the **molarity** of the new solution?

$$M_1 V_1 = M_2 V_2$$

$$(0.6) = M_2 (2.6)$$

$$(2.0) (0.30) = M_2 (2.6)$$

$$M_2 = 0.23 \text{ M}$$

- 2) Before an experiment, a scientist obtains 2.45 L of 1.25 M sulfuric acid (H₂SO₄) and dilutes it until the molarity drops to 0.80 M. What is the **volume** of the new solution?

$$M_1 V_1 = M_2 V_2$$

$$(3.06) = (0.8) V_2$$

$$(1.25) (2.45) = (0.8) V_2$$

$$V_2 = 3.83 \text{ L}$$

- 3) A solution of hydrochloric acid is diluted, bringing the volume from 2.0 L to 8.5 L. If the solution has a final molarity of 1.4 M, what was the **molarity before** dilution?

$$M_1 V_1 = M_2 V_2$$

$$M_1 (2.0) = 11.9$$

$$M_1 (2.0) = (1.4) (8.5)$$

$$M_1 = 5.95 \text{ M}$$

- 4) If a student adds 250 mL of water to a 300 mL solution of NaCl that has a concentration of 3.0 M, what will be the **molarity** of the final solution?

$$M_1 V_1 = M_2 V_2$$

$$(900) = M_2 (550)$$

$$(3.0) (300) = M_2 (550)$$

$$M_2 = 1.64 \text{ M}$$



- 5) A scientist adds 1.2 liters of water to a 5.0 liter solution, causing the molarity to become 0.4 M. What was the **molarity of the initial solution** before the dilution?

$$M_1 V_1 = M_2 V_2$$

$$M_1 (5.0) = 2.48$$

$$M_1 (5.0) = (0.4) (6.2)$$

$$M_1 = 0.50 \text{ M}$$

Solutions Molarity And Dilution Practice Answer Key

Clemens Wendtner



Solutions Molarity And Dilution Practice Answer Key:

The book delves into Solutions Molarity And Dilution Practice Answer Key. Solutions Molarity And Dilution Practice Answer Key is an essential topic that needs to be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Solutions Molarity And Dilution Practice Answer Key, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:
 - Chapter 1: Introduction to Solutions Molarity And Dilution Practice Answer Key
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 - Chapter 4: Solutions Molarity And Dilution Practice Answer Key in Specific Contexts
 - Chapter 5: Conclusion
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