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USER MANUAL



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web $q = 140 g \cdot 4.184 J / 160C = 9372 J$
 $16 J / 35.5 g \cdot 1 \text{ mole } KCN = 0.5453 \text{ moles}$
 $65.1 g \cdot 0.5453 \text{ mole} = 35.5 \text{ moles}$
 when a 15 g sample of SO_2 dissolves in 80 g of water in a calorimeter the temperature drops from 42°C to 15°C calculate the ΔH for the process
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