

NAME: _____

SECTION: _____

TIME: _____

Matching: Use each answer once only. Place the letter that best matches the description by writing the letter in the blank next to the name. (8 points each)

- | | | | |
|----------|---|--|------------------------|
| <u>1</u> | A | A type of covalent compound | A. None |
| <u>2</u> | B | A type of ionic compound | B. Amphoteric |
| <u>3</u> | C | Having shared electrons concentrated in the central atom | C. Amphiprotic |
| <u>4</u> | D | Forming giant lattices with different geometries | D. Polar bond strength |
| <u>5</u> | E | A lattice with the giant ionic lattice structure | E. Polar bond |
| <u>6</u> | F | Something whose properties can be distinguished | F. Amphoteric |
| | | | G. Amphoteric |
| | | | H. Amphoteric |
| | | | I. Amphoteric |
| | | | J. Amphoteric |
| | | | K. Amphoteric |
| | | | L. Amphoteric |
| | | | M. Amphoteric |
| | | | N. Amphoteric |
| | | | O. Amphoteric |
| | | | P. Amphoteric |
| | | | Q. Amphoteric |
| | | | R. Amphoteric |
| | | | S. Amphoteric |
| | | | T. Amphoteric |
| | | | U. Amphoteric |
| | | | V. Amphoteric |
| | | | W. Amphoteric |
| | | | X. Amphoteric |
| | | | Y. Amphoteric |
| | | | Z. Amphoteric |

Multiple Choice: Each part has one correct choice. (8 points each)

- 1 A. The number of lone pairs on the central atom in the following molecule is:
- a. 0
 - b. 1
 - c. 2
 - d. 3
 - e. 4
- 2 B. The number of lone pairs on the central atom in the following molecule is:
- a. 0
 - b. 1
 - c. 2
 - d. 3
 - e. 4

Assessment 4 2 Module 4 Investing Answers

Mario Herger



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