

# CHIMICA GENERALE

Ogni atomo/elemento è caratterizzato da un simbolo



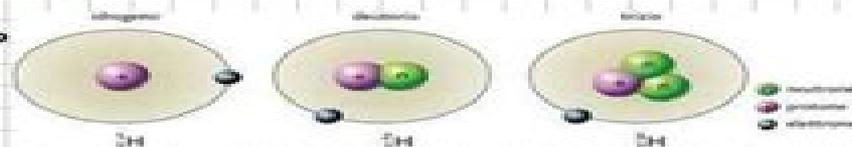
→ il numero di massa è dato dalla somma di protoni + neutroni (nucleoni)

→ il numero atomico corrisponde al numero dei protoni (uguale al numero di e<sup>-</sup>)

un nucleo con particolari valori di Z e A → **NUCLIDE**

**GLI ISOTOP:**

Atomi di uno stesso elemento che possono differire per il numero di neutroni. I diversi isotopi di un elemento hanno uguali proprietà chimiche, ma diverse proprietà fisiche.



→ **CATIONI:** quando ad un atomo vengono sottratti uno o più elettroni, si formano ioni carichi positivamente.

→ **ANIONI:** quando l'atomo assume uno o più elettroni si formano ioni carichi negativamente.

- contiene 7 periodi (orizzontali)  
- 18 gruppi (verticali) → nei gruppi gli elementi hanno caratteristiche chimiche simili

**TAVOLA DEGLI ELEMENTI:**

essi sono organizzati per massa atomica crescente.

gruppi	1A	2A	3A - 10A										11A	12A	13A	14A	15A	16A	17A	18A																
1	H	He																																		
2	Li	Be											B	C	N	O	F	Ne																		
3	Na	Mg	Al	Si	P	S	Cl	Ar	K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr										
4	K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr	Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe
5	Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe	Cs	Ba	La	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	
6	Cs	Ba	La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn				
7	Fr	Ra	Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Mn	Uu	Uu	Uu	Uu	Uu	Uu	Uu	Uu	Uu	Uu	Uu	Uu	Uu	Uu	Uu	Uu	Uu	Uu	Uu	Uu	

- **GRUPPO 1:** metalli alcalini (Li, Na, K, Rb, Cs)
- **GRUPPO 2:** metalli alcalino-terrosi (Be, Mg, Ca, Sr, Ba)
- **GRUPPO 18:** gas nobili, essi si combinano con pochissimi altri elementi. Sono tutti gas incolori e inodori. Sono poco reattivi.
- **GRUPPO 17:** alogeni. Si presentano con molecole biatomiche.

- **I METALLI:** sono solidi, dotati di lucentezza, malleabilità e duttilità; essi sono buoni conduttori di elettricità e calore.

- **I NON METALLI:** sono gas o solidi che non hanno caratteristiche metalliche.

- **SEMIMETALLI O METALLOIDI:** hanno caratteristiche intermedie tra i metalli e i non metalli.

# Appunti Di Chimica Generale Unife

**Lingsheng Yao**



## **Appunti Di Chimica Generale Unife:**

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## The Enigmatic Realm of **Appunti Di Chimica Generale Unife**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing short of extraordinary. Within the captivating pages of **Appunti Di Chimica Generale Unife** a literary masterpiece penned by a renowned author, readers attempt a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting effect on the hearts and minds of those who partake in its reading experience.

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