

Name \_\_\_\_\_

**MULTIPLE CHOICE.** Choose the one alternative that best completes the statement or answers the question.

- 1) According to kinetic-molecular theory, in which of the following gases will the root-mean-square speed of the molecules be the highest at 200°C?
- A) SF<sub>6</sub>  
 B) H<sub>2</sub>O  
 C) HCl  
 D) Cl<sub>2</sub>  
 E) None. The molecules of all gases have the same root-mean-square speed at any given temperature.

**SHORT ANSWER.** Write the word or phrase that best completes each statement or answers the question.

- 2) What two properties of real gases cause deviation from ideal behavior?

**MULTIPLE CHOICE.** Choose the one alternative that best completes the statement or answers the question.

- 3) A sample of a gas (5.0 mol) at 1.0 atm is expanded at constant temperature from 10 L to 15 L. The final pressure is \_\_\_\_\_ atm.
- A) 1.5                      B) 15                      C) 0.67                      D) 3.3                      E) 7.5
- 4) A sample of He gas (2.35 mol) occupies 57.9 L at 300.0 K and 1.00 atm. The volume of this sample is \_\_\_\_\_ L at 423 K and 1.00 atm.
- A) 0.709                      B) 57.9                      C) 41.1                      D) 81.6                      E) 1.41
- 5) The density of N<sub>2</sub>O at 1.53 atm and 45.2°C is \_\_\_\_\_ g/L.
- A) 0.388                      B) 2.58                      C) 9.99                      D) 1.76                      E) 18.2
- 6) The volume of a sample of gas (2.49 g) was 752 mL at 1.98 atm and 62°C. The gas is \_\_\_\_\_.
- A) NO<sub>2</sub>                      B) SO<sub>3</sub>                      C) SO<sub>2</sub>                      D) Ne                      E) NH<sub>3</sub>
- 7) What volume (mL) of sulfur dioxide can be produced by the complete reaction of 3.82 g of calcium sulfite with excess HCl (aq), when the final SO<sub>2</sub> pressure is 827 torr at 44°C?
- A)  $1.39 \times 10^{-4}$                       B) 0.106                      C)  $1.00 \times 10^{-3}$                       D) 761                      E) 578
- 8) The Mond process produces pure nickel metal via the thermal decomposition of nickel tetracarbonyl:



What volume (L) of CO is formed from the complete decomposition of 444 g of Ni(CO)<sub>4</sub> at 752 torr and 22°C?

- A) 255                      B) 20.2                      C) 63.7                      D) 11.0                      E) 0.356

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