

REPORT SHEET  
**Acid-Base Titration**

LAB  
**10**

**A. Concentration of Acetic Acid in Vinegar**

1. Brand Heinz Volume 5.00 mL (% on label) 5.0 %  
 2. Molarity (M) of NaOH \_\_\_\_\_ M

|                                 | Trial 1 | Trial 2 | Trial 3 |
|---------------------------------|---------|---------|---------|
| 3. Initial NaOH level in buret  | 0.1 mL  | 0.2 mL  | 0.1 mL  |
| 4. Final NaOH level in buret    | 39.6 mL | 41.2 mL | 40.1 mL |
| 5. Volume (mL) of NaOH used     |         |         |         |
| 6. Average volume (mL)          |         |         |         |
| 7. Average volume in liters (L) |         |         |         |

Show all calculations for the following questions.

8. Moles of NaOH used in titration \_\_\_\_\_ mole NaOH
9. Moles of  $\text{HC}_2\text{H}_3\text{O}_2$  neutralized by NaOH \_\_\_\_\_ mole  $\text{HC}_2\text{H}_3\text{O}_2$
10. Molarity of  $\text{HC}_2\text{H}_3\text{O}_2$  \_\_\_\_\_ M  $\text{HC}_2\text{H}_3\text{O}_2$
11. Grams of  $\text{HC}_2\text{H}_3\text{O}_2$  \_\_\_\_\_ g  $\text{HC}_2\text{H}_3\text{O}_2$
12. Percent (m/v)  $\text{HC}_2\text{H}_3\text{O}_2$  in vinegar \_\_\_\_\_ %  $\text{HC}_2\text{H}_3\text{O}_2$

# Acid Base Titration Lab Answers Chem Fax

**Y Pai**



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