

Texts in Statistical Science

Fundamentals of Mathematical Statistics



Steffen Lauritzen



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A CHAPMAN & HALL BOOK

Mathematical Statistics

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web write the oxidation and reduction reactions for each redox reaction the first one is done for you $23 \text{ Fe} + \text{CO}_2 \rightarrow \text{Fe}_2\text{O}_3 + \text{C}$
 oxidation $\text{CO} + \text{CO}_2 + 2\text{e}^-$ reduction $\text{Fe}^{2+} + 2\text{e}^- \rightarrow \text{Fe}$ $24 \text{ 3 Ag} + \text{Ni}^{3+} \rightarrow 3 \text{ Ag} + \text{Ni}$ oxidation reduction $25 \text{ Cu} + \text{Pb}^{2+} \rightarrow \text{Cu}^{2+} + \text{Pb}$ oxidation reduction 26 O_2

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 $\text{I}_2 + 2\text{O}_5 \rightarrow 2\text{IO}_3 + \text{O}_2$
 $5\text{CO} + \text{I}_2 \rightarrow 5\text{CO}_2 + \text{I}_2$
 $5\text{C} + \text{O}_2 \rightarrow 5\text{CO}_2$

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