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## Student Exploration: pH Analysis: Quad Color Indicator

**Vocabulary:** acid, acidic, alkaline, base, indicator, neutral, pH

**Prior Knowledge Questions** (Do these BEFORE using the Gizmo.)

1. **Acids** are substances that produce hydrogen ions ( $H^+$ ) when dissolved in water. Lemon juice is an example of an acid.

A. What does lemon juice taste like? sour

B. What does it feel like if lemon juice gets in your eye? burns

2. **Bases** are substances that produce hydroxide ions ( $OH^-$ ) when dissolved in water. Hand soap is an example of a base.

A. What does soap feel like? slippery, smooth

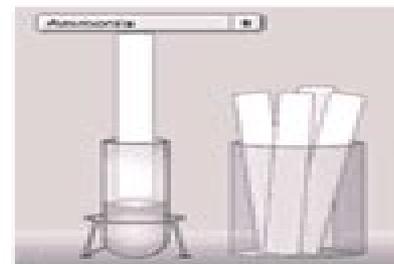
B. What does soap taste like? sour, bad

C. What does it feel like if soap gets in your eye? burns

### Gizmo Warm-up

The strength of an acid or base is measured on the **pH** scale. The term "pH" is short for "potential of hydrogen." It is a measure of how many excess  $H^+$  ions there are in a solution. The pH scale runs from 0 to 14, with 0 representing the highest concentration of hydrogen ions. **Acidic** substances have a pH below 7, while **alkaline** substances (bases) have a pH above 7. Pure water has a pH of 7 and is considered **neutral**.

The *pH Analysis: Quad Color Indicator* Gizmo™ allows you to find the pH of a variety of liquids. In the Gizmo, check that the **Substance in the tube** is **Ammonia**, and click **Test**.



1. **Indicators** change color in acids or bases. What are the colors of the pH paper?

orange, coral pink, blue, yellow

2. Compare the paper to the **pH color chart**. What is the pH of ammonia? 11

Is ammonia acidic or alkaline? alkaline

# Student Exploration Ph Analysis Gizmo Answer Key

**JA Banks**



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## **Table of Contents Student Exploration Ph Analysis Gizmo Answer Key**

1. Understanding the eBook Student Exploration Ph Analysis Gizmo Answer Key
  - The Rise of Digital Reading Student Exploration Ph Analysis Gizmo Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Student Exploration Ph Analysis Gizmo Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Student Exploration Ph Analysis Gizmo Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Student Exploration Ph Analysis Gizmo Answer Key
  - Personalized Recommendations
  - Student Exploration Ph Analysis Gizmo Answer Key User Reviews and Ratings
  - Student Exploration Ph Analysis Gizmo Answer Key and Bestseller Lists
5. Accessing Student Exploration Ph Analysis Gizmo Answer Key Free and Paid eBooks

- Student Exploration Ph Analysis Gizmo Answer Key Public Domain eBooks
  - Student Exploration Ph Analysis Gizmo Answer Key eBook Subscription Services
  - Student Exploration Ph Analysis Gizmo Answer Key Budget-Friendly Options
6. Navigating Student Exploration Ph Analysis Gizmo Answer Key eBook Formats
- ePub, PDF, MOBI, and More
  - Student Exploration Ph Analysis Gizmo Answer Key Compatibility with Devices
  - Student Exploration Ph Analysis Gizmo Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Student Exploration Ph Analysis Gizmo Answer Key
  - Highlighting and Note-Taking Student Exploration Ph Analysis Gizmo Answer Key
  - Interactive Elements Student Exploration Ph Analysis Gizmo Answer Key
8. Staying Engaged with Student Exploration Ph Analysis Gizmo Answer Key
- Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Student Exploration Ph Analysis Gizmo Answer Key
9. Balancing eBooks and Physical Books Student Exploration Ph Analysis Gizmo Answer Key
- Benefits of a Digital Library
  - Creating a Diverse Reading Collection Student Exploration Ph Analysis Gizmo Answer Key
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Student Exploration Ph Analysis Gizmo Answer Key
- Setting Reading Goals Student Exploration Ph Analysis Gizmo Answer Key
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Student Exploration Ph Analysis Gizmo Answer Key
- Fact-Checking eBook Content of Student Exploration Ph Analysis Gizmo Answer Key
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

#### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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